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Journal of the European Ceramic Society

journal homepage: www.elsevier.com/locate/jeurceramsoc



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Ceramics and ceramic coatings in orthopaedics

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ARTICLE INFO

Article history:
Received 20 April 2015
Received in revised form 28 July 2015
Accepted 29 July 2015
Available online 2 October 2015

Keywords: Bioceramics Ceramic coatings Implants Orthopaedics Arthroplasty

ABSTRACT

Advanced bioceramics have played integral roles in treatment modalities for damaged or diseased human joints and osseous defects. This paper reviews the uses and properties of ceramics and ceramic coatings variously employed as articulation devices in hip, knee, shoulder, and other joints, either as self-mated surfaces, or against polyethylene (both conventional and highly cross-linked versions), or for osseous-fixation as arthrodesis devices, bone scaffolds, and substitutes in the spine or extremities. The modern uses of oxide and non-oxide materials in these applications will be discussed, followed by an assessment and comparison of their mechanical and physicochemical properties. Recent developments in new bioceramic materials and composites along with advanced processing and testing methods are presented. Advanced bioceramics and coatings are expected to have increasing use in orthopaedics because of their unique combination and range of properties including strength and toughness, hardness and wear resistance, biocompatibility, bacteriostasis, and osseointegration.

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1. Introduction

Orthopaedic reconstructive surgery typically involves the use of prosthetic biomaterials to repair or replace damaged or diseased musculoskeletal tissue in order to relieve pain and restore function for patients suffering from bone fractures, tumors, congenital deformities, arthritis, osteoporosis, scoliosis, infection, and other etiologies coincident with an ageing population. While these procedures were envisioned by medical practitioners for a long time, it was not until the advent of safe anesthesia techniques, sterility, and improved biomaterials in the 20th century that prosthetic correction of bone defects and replacement of diseased or damaged joints became practical [1]. The pioneering work in total hip arthroplasty by Sir John Charnley in the 1960s [2,3] paved the way for reconstructive procedures in other areas, including spine, knee, shoulder, ankle, wrist, and phalanges. Worldwide, the number of people affected from bone and joint disorders was estimated to be 1.78 billion in 2012, with spinal and lower extremity disabilities being the most prevalent [4]. In 2010-2011, the number of musculoskeletal surgeries in the US alone was 3.79 million [5], with joint arthroplasty accounting for more than half of this total [6–10]. Further, from a global perspective, approximately 2.9 million joint replacement surgeries take place annually, including more than 1.4 million hip, 1.1 million knee, and over 100,000 shoulder replacements [11].

Ceramics and ceramic coated devices have been incorporated into various prostheses almost from inception either for osseousfixation or articulation devices. Current ceramics include alumina (Al₂O₃), zirconia (ZrO₂), zirconia-toughened alumina (ZTA), alumina matrix composites (AMC), alumina-toughened zirconia (ATZ), silicon nitride (Si₃N₄), and hydroxyapatite (HAp); whereas ceramic coatings encompass diamond-like carbon (DLC), titanium nitride (TiN), zirconium nitride (ZrN), titanium-niobium-nitride (TiNbN), calcium phosphates $(Ca_3(PO_4)_2)$, and also hydroxyapatite $(Ca_{10}(PO_4)_6 (OH)_2)$. Although not strictly regarded as a coating (but as a native oxide layer), another popular metallic-ceramic composite is an in situ grown monoclinic zirconia onto a zirconium niobium alloy (i.e., oxidized zirconia, OxZr). All of these materials have found various applications in orthopaedic surgery during the late 20th and early 21st centuries. The purpose of this review is to discuss the uses of these ceramics and ceramic coatings, provide comparative analyses of their physical and mechanical properties, present their in vitro and in vivo performance, and correlate these to physicochemical observations.

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2. Modern uses of orthopaedic ceramics and coatings

An excellent review of the early history of arthroplasty including the use of ceramics was provided by Steinberg et al [1]. This section will not retrace their effort but instead focus on developments during the past century. There has been explosive growth in the science and application of biomaterials in joint replacement during the past 50 years. In fact, total hip arthroplasty has been referred to as the most successful operation of the 20th century [12]; and its success has led to the development of arthroplastic procedures for other human joints. Ceramic materials have been integral to this evolution. They were first considered for use as prostheses in the 1920s when hip reconstructive procedures were devised using hollow hemispherical molded glass as femoral heads. However, these devices lacked sufficient strength and toughness to withstand acetabular loads and many fractured in vivo [13]. A nearly 50 year hiatus ensued during which physicians turned principally to metals and polymers. However, in the latter third of the 20th century, orthopedists returned to ceramics because of favorable properties such as biocompatibility, strength, hardness, wear, and corrosion resistance. This period saw the introduction and proliferation of alumina, hydroxyapatite, zirconia, silicon nitride, ceramic-coated metals, and zirconia-toughened alumina. For an in-depth treatise on the physicochemical characteristics of most of these biomaterials, the interested reader is referred to a recently published work by Pezzotti [14].

2.1. Alumina

Interest in polycrystalline bioceramics started in 1970 when Dr. Pierre Boutin implanted the first hip prostheses with Al₂O₃ ceramic bearings [15,16]. His success led to a string of ceramic biomaterial innovations, with patents appearing shortly thereafter claiming various ceramics, including Al₂O₃, as viable orthopedic materials [17-20]. Recognizing a potential for limb reconstruction and correction of skeletal defects in veterans, the US Defense Department investigated and demonstrated the usefulness of porous Al₂O₃ as a bone scaffold in 1972 [21], and a similar Scandinavian study in 1973 established its osseointegration effectiveness [22]. Al₂O₃ has been the most widely used structural ceramic in total hip arthroplasty (THA) for the past approximately 50 years, typically in articulation against polyethylene [23] or itself [24,25]. Its success in THA led to its adoption in other joints including knee [26,27], elbow [28], ankle [29], wrist [30], phalanges [31], spine [32,33], and also bone reconstruction [34]. Industrial specifications for its composition, processing, and properties have been issued [35,36]. It has been repeatedly shown to be biocompatible [16,37], possessing high hardness and a low coefficient of friction, all of which significantly reduce the occurrence of wear debris [38]. However, it also has relatively modest fracture strength and toughness, both of which increase the risk of brittle failure [39,40]. Yet, enhanced reliability has been realized with three generations of processing improvements, including increased raw material purity, refinement of its microstructure, use of hot- isostatic pressing (HIP), and inclusion of post-manufacturing proof testing [41].

2.2. Hydroxyapatite and calcium phosphates

Apatite-based ceramics were investigated and introduced in the 1960–70s concurrent with Al $_2$ O $_3$ [42], although their therapeutic bone remodeling benefits have been known for centuries [43]. β -Tri-calcium phosphate (Ca $_3$ (PO $_4$) $_2$, TCP) and its close chemical cousin, calcium hydroxyapatite (Ca $_1$ 0(PO $_4$) $_6$ (OH) $_2$, HAp), in granular, porous and dense forms, were suggested as corrective therapies for various bone defects, such as tooth repair and replacement, alveolar ridge augmentation, maxillofacial reconstruction,

ossicles bone repair, or as burr-hole buttons, orbital implants, spinal fusion devices, and bone scaffolds [44–46]. Their early use in dental procedures resulted in marked improvement in osseointegration [47]. This success provided incentives to other researchers to apply hydroxyapatite coatings onto femoral stems and cups in total hip arthroplasty. Clinical trials of coated hip stems were initiated by the mid-1980s [48], with follow-up on a number of these studies now exceeding 20 years [49-52]. Today HAp is the most used non-structural ceramic in joint arthroplasty [53]. Unfortunately, it cannot be employed in load bearing devices due to its extreme brittleness and poor strength. Its usefulness is solely for the promotion of new bone growth. HAp is considered osteoconductive similar to autologous or allograft bone. When exposed to physiologic fluids, it releases ions that stimulate bone formation and bone bonding, which aids in the stabilization and fixation of implants possessing porous ingrowth surfaces. Although porous metals, such as titanium alloys, are capable of similar bone ongrowth or ingrowth even without hydroxyapatite coatings, HAp accelerates the appositional process, and is used clinically. The original (and still preferred) method of applying hydroxyapatite is by thermal spraying of synthesized powders at temperatures between 15,000 K and 30,000 K to form thin $(50-150 \,\mu\text{m})$ surface coatings, with roughened or porous titanium being the preferred metallic member [54,55]. Melting of the hydroxyapatite occurs within the plasma stream, and fortuitously, when the particles impact on the cooler metallic substrate, a range of disordered calcium phosphate phases are formed possessing differing biologic dissolution kinetics, which favor both short- and long-term osseointegration. These phases include predominately amorphous calcium phosphate (ACP), α - and β -tri-calcium phosphates (TCP), and hydroxyapatite (HAp). The reported relative biological solubility of these compounds is as follows: ACP $\gg \alpha$ - or β -TCP \gg HAp [55]. Their dissolution rate is not only related to crystal chemistry, but also affected by physical parameters such as the size of melted splats, the amount of porosity within the coating, the presence of thermally induced cracks, pits or other defects, and the coating's crystallinity. Reportedly, initial bone growth occurs at an enhanced rate when the coating contains a higher percentage of the amorphous phase [55]. Its dissolution increases the local concentration of calcium and phosphate ions followed by precipitation of a calcium carbonaceous phosphate with a composition similar to native bone. These precipitates attract osteoblasts inducing bone growth towards the implant. Concurrently, osteoclasts remodel native bone surrounding the implant according to applied physiologic stresses. The osteoclast cells dissolve the carbonate precipitates as new native bone grows into fissures, pits, and defects of the slower resorbing HAp during post-operative periods. Poor fixation and implant loosening may occur if the applied coating has too much ACP or TCP. High concentrations of these two phases results in rapid bone formation at the expense of longer-term osseointegration. Conversely, an insufficient amount of these compounds will fail to attract the necessary osteoblasts to initiate fixation; or the higher concentration of HAp can occlude the porosity of the underlying metallic surface thereby failing to achieve adequate apposition of the implant and native bone [56,57]. Herein lies the key concern and controversy of HAp coated implants. Resorption kinetics govern implant fixation. Thermal spraying conditions have been experimentally determined to achieve appropriate concentrations of ACP, TCP, and HAp. The HAp feedstock composition, purity, particle size, gas- stream temperature and its composition, and substrate temperature are among the important experimental parameters. Recent work has also explored the preparation and characterization of biomimetic and nano-hydroxyapatite compositions with the hope that they might also lead to improved implant fixation [58-60]. Although not indispensable in arthroplasty surgery, HAp and its associated calcium phosphates are useful for promoting bone ingrowth and accelerating the biologic anchoring of joint prostheses to host bone. Their effectiveness for this purpose continues to be robustly demonstrated and reviewed [53,61–63,656].

2.3. Zirconia

Continuing concerns over in vivo fractures of Al₂O₃ femoral heads and liners in total hip arthroplasty in the 1980s led to the introduction of zirconia ceramics in 1985 [64], including partially stabilized zirconia using magnesia (Mg-PSZ) and a yttria-doped composition known as Tetragonal Zirconia Polycrystals (Y-TZP) [64–66]. These ceramics represented a new generation of implants. They were biocompatible [37] and had approximately twice the strength and toughness of Al₂O₃ [66]. They also had exceptionally low coefficients of friction and lower wear on polyethylene. They promised to provide a new level of care and safety over metallic and Al₂O₃ implants [64,67–69]. Their improved mechanical properties were achieved by taking advantage of a polymorphic phase transformation that occurs in doped zirconia known as phase transformation toughening [70]. This mechanism involves the meta-stabilization of a high-temperature tetragonal phase through the incorporation of magnesia, yttria, or other dopants. In the presence of an advancing crack, the metastable phase transforms to its stable monoclinic polymorph with an accompanying volume increase of \sim 4–5%. The sudden volume change exerts compressive forces on the crack tip, thereby slowing or arresting its propagation. Mg-PSZ implants entered the marketplace and an industrial standard for their composition, processing, and properties was created [71]. However, the strength of Mg-PSZ was inferior to Y-TZP, so, considerable physician interest and rapid uptake of Y-TZP ceramics occurred after its clearance by US and EU regulatory bodies. Over 600,000 Y-TZP femoral heads were implanted between 1985 and the turn of the century largely based on bench test data, with little clinical evidence as to the material's longer-term effectiveness [72,73]. There were initially few reported in vivo failures [69], but a change in manufacturing methods (from batch to tunnel furnace) by the leading vendor of Y-TZP in 1998 resulted in a significant increase in fractures beginning in 2000, many occurring only months after implantation [73]. Independent investigations by a number of scientific groups in cooperation with regulatory agencies highlighted the material's inherent instability [72-74]. After more than a decade of use, Y-TZP was withdrawn from the US and EU markets in 2001 and its industrial standard was subsequently abandoned [75]. Despite its improved initial strength, Y-TZP possesses one critical weakness. It may spontaneously transform to its stable monoclinic form under in vivo conditions. This effect has become known as low-temperature hydrothermal degradation (LTD) [76]. The transformation results in increased implant surface roughness, enhanced wear, weakening of the material, and eventual fracture [72–74,77,78]. Despite its withdrawal from EU and US markets, Y-TZP femoral heads are still available for implantation in Asia, and recent physical chemistry analyses suggest that the unconstrained transformation experienced by the original EU manufacturer may be limited and controlled through improved processing and careful assessment before clinical use [79]. Y-TZP is also extensively used and preferred as esthetic and functional dental prostheses [80], although concerns have also been raised about its longevity in this application due to low-temperature hydrothermal degradation [81]. LTD is considered the Achilles' heel of zirconia with its occurrence being highly dependent on its microstructure; therefore, the overall process must be strictly controlled and any change in one of the process steps must be carefully assessed. In other, simple terms, Y-TZP may perform better than Al₂O₃ but is less robust as far as industrial variations are concerned.

2.4. Silicon nitride

Silicon nitride (Si₃N₄) is another high-strength and tough industrial ceramic which is also considered to be a viable implant material [82-84]. In 1986, it was first utilized in a small clinical trial in Australia (30 patients) to determine its capability as an arthrodesis device in the lumbar spine [85]. This medical study happened to coincide with massive investments by the US Government in the development of this new ceramic for aerospace, automotive. and industrial applications [86–90], although no funding was ever channeled to medical devices at that time. Over two decades passed before Si₃N₄ became a commercially available biomaterial in the US and EU, partially thanks to a small grant from the US National Institutes of Health [91]. Since 2008, it has been used as a fusion cage for arthrodesis of the cervical and thoracolumbar spine [92]; and to date, approximately 25,000 Si₃N₄ spinal fusion cages have been implanted with few adverse reported events [93]. Extensive testing of silicon nitride has indicated its potential as an articulation member as well, but it has yet to be cleared by regulatory agencies for this purpose [82,94]. Two industrial standards have been adopted for its composition, processing, and properties [95,96]. They are appropriate for its use as a biomaterial, but need to be augmented with provisions that address biologic issues. Si₃N₄ has been shown to be biocompatible, possessing favorable cell interaction characteristics [83,84,97-102]. These and other studies indicate that porous or unpolished Si₃N₄ osteointegrates with adjacent bone [102-106] and exhibits bacteriostasis [106,107]. In its dense and polished form, it has been shown to produce implants with exceptionally low wear rates [108–118]. Its strength and toughness are comparable to the best values obtained for other polycrystalline ceramics [82]. Si₃N₄ derives its strength and toughness through microstructural engineering and not from phase transformation. Similar to Al₂O₃, Si₃N₄ exists as an irreversibly stable phase at room temperature; but unlike Al₂O₃, its microstructure is composed of asymmetric needle-like interlocking grains surrounded by a thin (<2 nm) refractory grain-boundary glass [119]. This unique structure provides Si₃N₄ with exceptional strength and toughness. An advancing crack must navigate a tortuous high energy path through the ceramic, and bridging grains within the crack wake restrict its continued propagation [120-122]. Si₃N₄ can not only be produced as a polycrystalline ceramic, but it can also be applied as a wear resistant coating using physical vapor (PVD) or chemical vapor deposition (CVD) methods [123-127]. Si₃N₄ is the only ceramic material being considered thus far for both bulk and metal-coated orthopedic applications. Although bulk and coated silicon nitrides are chemically similar, the coated material is amorphous or nano-crystalline and non-stoichiometric. It can be either silicon or nitrogen rich and may also contain other elements, either by design or as impurities, such as carbon, oxygen, or fluorine. Mechanically, it does not have the same strength or toughness as the polycrystalline material, but can have comparable hardness.

2.5. Ceramic coatings for bearing applications

Other ceramic coatings for medical applications predated silicon nitride. The first of these was PVD deposited titanium nitride (TiN) on titanium. It was developed and introduced into the market-place for total joint arthroplasty beginning in the late 1980s [128] with formal clinical investigations beginning in 1990 [129–131]. At about this same time, another hard coating, diamond-like carbon (DLC), was applied to titanium implants for shoulder, ankle, and knee arthroplastic devices in Europe. However, no pre-operative *in vitro* testing of these DLC coated implants was ever conducted. Neither were there any clinical data nor regulatory oversight because the manufacturer failed to obtain the necessary clearances prior to marketing these devices. Consequently, these products fared poorly

and were quickly removed from distribution [132]. The first and only legitimate premarket approved clinical trial of DLC coated devices was initiated in 1993 [133]. Other hard ceramic coatings were subsequently introduced, including zirconium nitride (ZrN) on a cobalt chromium (CoCr) substrate [134] and titanium niobium nitride (TiNbN) also on CoCr [135,136]. These single or mixed nitride coatings are nano-crystalline and have thicknesses in the range of 1 µm to 15 µm. All have been shown to be biocompatible [123-127,136-139], and they are currently used or considered for use as hip resurfacing implants, for total hip arthroplasty, or as femoral knee components. In 1997 another new metal-ceramic composite was introduced via clinical trials [140,141]. Known by the trade name of Oxinium[®] [142], it is not technically a coating. It consists of a \sim 5 μ m thick monoclinic ZrO₂ ceramic layer which is grown in situ on a Zr-2.5Nb alloy. However, like coatings, it combines the strength of a metal core with the wear resistance of a ceramic surface. This composite has been shown to be biocompatible [143] and is currently successfully used in both hip and knee arthroplasty [144,145]. In principle, ceramic coatings appear to be excellent solutions for three prevalent problems in articulating prostheses. First, they eliminate brittle fracture because of their high toughness metallic substrates. Second, in comparison to metal implants, they minimize wear debris due to the high hardness and abrasion resistance of the outer ceramic layer. Third, they place a barrier between the metallic implant and human bone or tissue, which lessens allergic reactions from soluble metal ions (Co, Cr, and particularly Ni) for hypersensitive patients [146]. However, there are concerns with all thin ceramic coatings or layers. Their fabrication processes can generate high surface stresses, which challenge adhesion to their respective substrates. Thermal expansion differences can exacerbate these stresses. Also, in vivo scratching, pitting, or even delamination of these coatings have been observed [147–149], leading to increased surface roughness and excessive wear of the coating, the underlying metal, or the counterface material [150,151].

2.6. Zirconia toughened alumina

A group of new ceramics containing mixtures of Al₂O₃ and ZrO₂ were brought to US and EU markets beginning in about 2000, almost simultaneous with the withdrawal of Y-TZP [152]. They were introduced in an attempt to overcome the hydrothermal deficiencies of Y-TZP and the relatively modest strength of Al₂O₃. At one end of the composition spectrum are zirconia-toughened alumina (ZTA) [153,154] and an alumina matrix composite (AMC) [152]. ZTA is a fully dense ceramic consisting typically of about 7-25 wt.% unstabilized ZrO₂ or Y-TZP incorporated into an Al₂O₃ matrix [155,156]. Alternative ZTA materials have also been developed using cerium oxide as the stabilization additive instead of yttria [157-159]. AMC became a special ZTA composition consisting of 24 wt.% ZrO₂ (i.e., 17 vol.%) with 1.7 wt.% mixed oxides (Y₂O₃, Cr₂O₃, and SrO), the balance being Al₂O₃ (74 wt.%) [160]. Since its introduction in 2000, AMC has become the most popular orthopaedic ceramic for articulation implants used in the world today. It is commercially known as BIOLOX® delta [160]. An industrial standard for its composition and testing has been prepared and published [161]. At the opposite end of the spectrum, but still consisting of Al₂O₃ and ZrO₂, is alumina-toughened zirconia (ATZ) [162]. ATZ is also a completely dense ceramic composed of 80 wt.% Y-TZP and 20 wt.% Al₂O₃ [162]. As might be expected, this broad range of Al₂O₃-ZrO₂ mixtures resulted in marked differences in properties, but all are superior to both pure Al₂O₃ and Y-TZP. All have been tested and determined to be biocompatible [163,164]. They have excellent strength, toughness, and hardness [154,165,166], and are highly abrasion [167] and wear resistant [168–174]. Clinical studies using these devices have been on-going since their introduction with acceptable overall outcomes [172,175–178]. However, they all achieve their enhanced mechanical properties through transformation toughening, and may be susceptible to *in vivo* hydrothermal instability, although its effects are retarded due to the presence of the non-transforming Al₂O₃ phase.

Undoubtedly, there will be other ceramics and coatings considered for arthroplasty over the coming decades as scientists and engineers attempt to match specific material properties or characteristics to existing or new orthopaedic applications. However, in learning from past experience, they will likely be introduced gradually [179], and only after significant scientific *in vitro* and *in vivo* scrutiny.

In the following sections, a comparative review of the physical and mechanical properties of these ceramics and coatings will be provided and correlated with available *in vitro* and *in vivo* data for their specific applications. The physicochemical stability of each material will follow, along with a discussion of degradation mechanisms. A summary of important findings and future directions for research and product development will be given.

3. Properties and performance

This section is devoted to a comparative discussion of ceramic material properties, bio-stability, in vitro, and in vivo performance as they relate to specific applications. Certain properties are unavailable for some materials, particularly the coatings; and the range of applications precludes direct comparison of some performance characteristics. In addition, differences in testing methods used in assessing properties make precise comparisons problematic. Nevertheless, given in Tables 1a and 1b are representative values for the physical and mechanical properties of polycrystalline ceramics and ceramic coatings contained within this review, respectively. Two medical grade alloys (CoCr and Ti6Al4V) and polyetheretherketone (PEEK) are included in Table 1a for comparison purposes. The data provided in these tables were compiled from peer-reviewed literature cited in this review. It is important to note that these published values are not considered material specifications. ASTM and/or ISO standards have been prepared for the bulk ceramics [35,36,71,75,95,96,161] and key specifications from these standards are shown in Table 2. These specifications reasonably represent the properties that orthopaedic implants must meet when they leave their respective manufacturing plants. A perusal of this information provides an interesting contrast between reported data from technical literature and actual specifications from the standards. They indicate considerable equivalence for bioceramics used in similar applications. Standard specifications do not exist for the hard ceramic coatings or oxidized zirconium, although processing and characterization techniques have been published by ASTM and ISO organizations [180,181].

Parameters considered important for total joint arthroplasty are density, grain size, flexural and compressive strength, Weibull modulus, fracture toughness and slow crack growth (SCG), hardness and wear resistance, biocompatibility, corrosion, and low-temperature hydrothermal degradation (LTD). Additionally, for bioceramics used as arthrodesis devices or for bone fixation, such as HAp, Al $_2$ O $_3$, or Si $_3$ N $_4$, the ceramic's hydrophilicity, bacterial resistance, and osseointegration ability are important characteristics. Other properties such as elastic modulus, Poisson's ratio, thermal expansion, and conductivity are specific to each material and not directly comparable, although they may play important roles in performance. All are included in Tables 1a and 1b for the purpose of providing a reasonable reference source for these bioceramics and coatings.

Table 1aPhysical and mechanical properties, and performance of biomaterials.

Property or performance	Units	Alumina	Zirconia		Zirconia-alur	mina composites	5	Silicon nitride	Cobalt chromium	Ti6Al4V	PEEK	Cortical bone
Composition or designation	NA	Al ₂ O ₃	Mg-PSZ	Ce- or Y-TZP	m-ZTA	AMC	ATZ	Si ₃ N ₄	ASTM F799	ASTM F136	ASTM F2026	Collagen, proteins, HAp
Density	g/cc	3.98	5.65-5.77	6.00-6.05	4.25	4.37	5.51	3.22-3.35	8.29-8.50	4.43-4.50	1.29	1.5-2.0
Grain size	μm	<1.8	50 Equiaxed	0.1-0.6	0.4-0.7	0.54	0.4	0.5×5.0	~62	~10 × 60 Lamellar	NA	NA
		Equiaxed	1	Equiaxed	Equiaxed	Equiaxed	Equiaxed	Equiaxed	Equiaxed			
lexural or tensile	MPa	400-580	450-700 Flexural	700-1500	700- 1248	1250-1400	755–1163	800-1100	827 Tensile	860-970 Tensile	170	90-228
trength		Flexural		Flexural	Flexural	Flexural	Flex./biaxial	Flexural			Flexural	Flexural
Compressive	MPa	4100-5000	2000-3000	2000–2200	4000-4500	4300	~2600	4000	600-1800	800-970	118	150-260
trength		1100 0000	2000 3000	2000 2200	1000 1000	1300	2000	1000	000 1000	000 070	110	70-110 \perp
lastic modulus	GPa	380	200-250	210-223	340-390	358	240-250	296-313	197-210	105-120	4	7.5–25.8
idstic inoddias	Gru	300	200 230	210 223	310 330	330	2 10 250	230 313	137 210	103 120	•	5-20 ⊥
oisson's ratio	NA	0.23	0.30	0.30-0.33	~0.24	0.24	~0.28	0.27	0.27-0.32	0.31-0.34	0.4	0.19- 0.48
Veibull modulus	NA	5-29	22	7-87	NA	10-15	6–17	8-53	NA	NA	NA	NA
racture toughness	MPa m ^{1/2}	3.3-4.2	2.9–16.0	4.5-20.0	>4.1	6.4-8.5	8.0–12.0	4.4–15.0	50-100	46.3-93.3	7.6 kJ/m ²	1.0-5.0
racture tougimess	IVIF d III	3.3-4.2	2.9-10.0	4.3-20.0	×4.1	0.4-6.3	0.0-12.0	4.4-13.0	30-100	40.3-33.3	Impact Test	3.0-20.0
atigue resistance		0.52-0.84	0.45-0.90	0.37-0.92	NA	0.67	NA	0.50- 0.97	0.14-0.36	0.10-0.40	0.53-0.62	0.30-20.0
		0.52-0.64	0.45-0.90	0.57-0.92	INA	0.67	INA	0.50- 0.97	0.14-0.50	0.10-0.40	0.33-0.62	0.30-0.83
TH/K _{IC}	NIA	D	D	D	D	D	D	D	Manustra at	D	Manufact	D
iocompatibility	NA	Pass	Pass	Pass	Pass	Pass	Pass	Pass	Marginal	Pass	Marginal	Pass
urface phase	%	100%	42-54% t-ZrO ₂	65-95%	83-93%	58-90%	95-99%	100%	NA	Mixture of $\alpha \& \beta$ Ti	Amorphous	Collagen
Composition		α -Al ₂ O ₃		t-ZrO ₂	t-ZrO ₂	t-ZrO ₂	t-ZrO ₂	β -Si ₃ N ₄			&	and HAp
		6. 11							0.11	0.11	crystalline	
TD susceptibility	NA	Stable	Marginal	Metastable (Y-TZP; marginal (Ce-TZP)	Stable	Marginal	Metastable	Stable	Stable	Stable	Stable	NA
lardness	GPa	18.0-23.0	10.0-12.0	11.0-12.5	15.7-20.8	19	13.7-15.0	15.0	3.0-4.0	2.8-3.3	99 Rockwell M	0.68-0.78 0.46-0.57
Vear rate PE	mm ³ /MC	20-58	- 1.8-5.1 NA	11-63	NA	1-20	17-32	17-25	14-201	NA	NA	NA
HXLPE		0.0-6.9		5.0-6.0		0.1-4.4	5.6-6.1	3.7-6.3	0.0-11.7			
lard-on-hard		0.02 - 4.71		Catastrophic		0.00-0.45	0.02-0.06	0.18-0.98	0.18-25.00			
hermal expansion	10 ⁻⁶ /°C	8	7-10	11	~8	8.1	~10	2.0-4.6	7.32	8.5-9.7	47	22.0-32.4
oefficient	,											
'hermal	W/m K	30	2	2-3	~17	17	~6	30-50	12.7	6.7-7.0	0.29	0.41-0.63
onductivity	***	30	-	2 3	• *	• •	· ·	30 00		017 710	0.20	0.11 0.03
K-ray	NA	Radiolucent	Opaque	Opaque	Opaque	Opaque	Opaque	Radiolucent	Opaque	Opaque	Transparent	Radioluce
adiolucency	1471	Rudioideein	Opuque	Opuque	Opuque	Opuque	Opuque	Radioraceire	opuque	Opuque	Transparent	Radiorace
essile water	Degree (°)	50-72	79	82	90	90	90	40-70	55-93	76	95	NA
ontact angle	Degree ()	30-72	73	02	30	30	30	40-70	33-33	70	33	1471
Bacteriostatic	⊕ = excellent;	0	+	+	NA	+	NA	⊕	8	+	Х	NA
apabilities	- ,	0	•		INU		IAU	Φ	⊗	'	^	11/1
*	+=good;				NIA		NIA					
Osseointegration	⊘ = fair;	\oslash	+	+	NA	+	NA	\oplus	\otimes	+	X	\oplus
ibility	\otimes = poor;											
	x = very poor											

NA = not applicable or not available. MC = Million cycles.

Table 1bPhysical and mechanical properties, and performance of biomaterials.

Property or performance	Units	Titanium nitride	Diamond-like carbon	Zirconium nitride	Titanium-niobium nitride	Oxidized zirconium	Hydroxyapatite
Composition or designation	NA	TiN	DLC	ZrN	TiNbN	Ox-Zr	ASTM F1609
Density	g/cc	4.87-5.22	0.90-3.20	7.09	~5.69	5.84	2.55-3.21
Grain size	μm	30-300 nmColumnar	Amorphous 2-25 nm	10-30 nmNanocrystals	10-30 nm Nanocrystals	$40 \times 200 nm$	0.4-100 μm splats
Adhesion or bond strength	MPa	10–60 N L _C Adhesion	35–160 NL _C Adhesion	24–60 N L _C Adhesion	83 N L _C Adhesion	35 N L _C Adhesion	39–89 Bond 25–60
Compressive strength	MPa	400-5500	NA	NA	NA	~2000	102–1000
Elastic modulus	GPa	402-550	110-900	175–395	200-600	200	3,2-122 coat vs. bulk
Poisson's ratio	NA	0.21	0.17-0.20	0.19	~0.20	0.34	0.11-0.27
Weibull modulus	NA	5–18	6–12	NA	NA	NA	2–19
Fracture toughness	$MPa m^{1/2}$	0.7-12.4	1.6-5.1	2.3-7.5	NA	2.2–2.8	0.5-1.2
Fatigue resistance		NA	NA	NA	NA	NA	0.61
K _{I0} /K _{IC} Biocompatibility	NA	Pass	Pass	Pass	Pass	Pass	Pass
Surface phase composition	%	TiN nanocrystals	Amorphous	ZrN nanocrystals	TiN, NbN nanocrystals	95% m-ZrO ₂ 5% t-ZrO ₂	ACP, TCP, HA
LTD Susceptibility	NA	Stable	Stable	Stable	Stable	Stable	Purposely degradable
Hardness	GPa	33-56	14.5-80.0	14.0-31.0	14.0-24.5	12.0-14.0	3.0-9.0
Wear rate PE	mm ³ /MC	21	28-67	NA	NA	-	NA
HXLPE	mm /wic	NA	2.8	3.5	NA	0.2-1.7	NA
Hard-on-hard		NA	NA	NA		NA	
Thermal expansion coefficient	10 ^{−6} /°C	7.4-9.2	2.3	5.9-7.2	~7.4–9.2	7–10	11.6–14.2
Thermal conductivity	W/m°K	11.9	0.2-30	20	~12–14	2–3	1.1-1.2
X-ray Radiolucency	NA	Opaque	Opaque	Opaque	Opaque	Opaque	Radiolucent
Sessile water contact angle	Degree (°)	31 - 69	55–71	89	73–75	71	34–39
Bacteriostatic capabilities	⊕ = excellent; + = good;	+	+	\oplus	NA	+	+
Osseointegration ability	<pre>> = good,</pre>	+	+	⊕	NA	NA	\oplus

NA = not applicable or not available. See text for discussion of adhesion and bond strength differences.

Table 2 ASTM or ISO specifications for biomaterials

ASTINI OI ISO SPECIII	ASTIM OF ISO SPECIFICATIONS FOR DIVINALEITAIS.	dis.									
Property	Al ₂ O ₃	Al ₂ O ₃ ISO6474-1	Mg-PSZ	Y-TZP	ZTA, AMC	Si ₃ N ₄	Si ₃ N ₄	CoCr	Ti6Al4V	PEEK	
	ASTM		ASTM	ASTM	ISO	ASTM	ISO	ASTM	ASTM	ASTM	
	F-603		F-2393	F-1873	6474-2	F-2094 ^a	26602ª	F799	F136	F2026	
Chemical	>99.5	>99.7	> 99.8	>99.0	>99.8	>97.0	NS	NA	>99.3	NA	
pullty (%) Density (g/cc	>3.93	≥3.94	>5.80	>6.00	≥4.31		3.0–3.6 NS	NA	NA	1.28-1.32	
and %)	>98.6		>98.8	>98.4	>98.6						
Grain size (μm)	<4.5 ≤4.5	≥2.5	NS	≥0.6	$Al_2O_3 \le 1.5$ $ZrO_2 < 0.6$	NS	NS	<64		Ϋ́	
Flexural strength	>400	> 500	>600	>800	>750		>760	827 (YS)	760 (YS)	110	
(MPa) ^a								1172 (TS) ^b	825 (TS)		
Weibull modulus	8	8 ^I	>10	NR	8	>12	>12	N	N	NA -3.	RI
Fracture toughness	NS	≥2.5	NS	NS	>3.5	>6.0		NA	NA		McEnti
(Mram''z) Hardness (GPa)	>18	>18	>10	≥12	>15.5	≥15	≥14.2	> 3.3	NA	ΝΑ	re et
Elastic modulus (GPa)	>380	>380	>180	>200	≥320	270-330	270–330	NA	NA		ai / lo
NC - Not cooping											n

NS = Not specified.

³ Two flexural strength determination methods are allowed within ASTM F-2094 and ISO-26602: For 3-point bending, the specification is \$900 MPa versus \$760-765 MPa for 4-point bending. ASTM and ISO specifications for the other bioceramics of Table 5 utilize 4-point bending.

(YS) = yield strength; (TS) = ultimate tensile strength Impact strength, notched izod method, (J/m).

3.1. Density

Although absolute density values (in g/cc) are not directly comparable for bioceramics due to chemical composition differences, attainment of near theoretical values (in %) provides insight with respect to each material's capability. Today, all ceramics for articulation purposes are densified using hot-isostatic pressing, and densities of greater than 98% are routinely achieved. HIPing results in substantial improvements in strength for these materials [16,182-185]. As an example, Garino et al. reported that HIPing of Al₂O₃ was implemented as part of a number of process improvements resulting in about a 3% change in mass density (i.e., from 3.86 g/cc to 3.96 g/cc) and a 45% increase in flexural strength (i.e., 400–550 MPa) [186]. Hot-isostatic pressing has since become a standard manufacturing practice for bulk ceramics used in structural applications [187]. The data presented in Table 1a and specifications shown in Table 2 demonstrate that all of these ceramics are essentially equivalent (within about 1.5%) with respect to their percentage of theoretical density.

Densities for ceramic coatings vary based on composition and application. Compiled values are provided in Table 1b. PVD coatings are typically dense amorphous or nano-crystalline structures [188]. Density determinations are difficult due to their thin nature, but measured values appear to approach theoretical limits in defectfree regions [189]. However, entrained imperfections, such as pits and micro-droplets, negatively impact average values. Oxidization of Zr-2.5Nb alloys (OxZr) produces a uniformly dense "blue-black" nano-crystalline scale. Nevertheless, its thickness is limited to about 5 µm because a porous "white" transitional oxide occurs beyond this point due to high compressive growth stresses [190]. HAp coatings are purposely engineered to be porous for osseous fixation. Porosity in thermally sprayed HAp coatings is governed by plasma conditions [191] and can vary from about 2% to greater than 10% [192,193]. A broad range of porosity can also be engineered into most ceramics for use as arthrodesis devices, scaffolds, or bone substitutes [21,105,194–197]. In summary, the amount of porosity within an implant is an important design parameter. Low porosity implants with concomitant small flaw populations are essential for load bearing applications whereas high porosity or mixed porosity devices are required for bone fixation.

3.2. Grain size and morphology

Grain size is an important parameter governing mechanical properties. The grain size values shown in Table 1a and the specifications of Table 2 are indicative of the control that is required in processing ceramics. Smaller grains are essential for maintaining a reasonable level of fracture strength in Al₂O₃. Minimizing grain size is the only method of improving the performance of devices made from this material [40,198]. While larger grained microstructures in Al₂O₃ resist crack propagation due to grain bridging and pull-out (which effectively increases local fracture toughness) [199,200], these large grains themselves can become inherent strength limiting flaws. Consequently, smaller grains are preferred. One approach to designing brittle ceramics for structural applications (as has been done for Al₂O₃) is to reduce the probability of failure through grain-refinement and flaw elimination, both of which increase fracture strength and improve Weibull modulus (i.e., less variability in strength). Provided in Fig. 1(a) is a microstructural example of a biomedical Al₂O₃ [154].

Unlike Al $_2$ O $_3$, Mg-PSZ has a unique microstructure. Its average grain size is large – 50–60 μm – and is composed of a cubic- ZrO $_2$ matrix embedded with metastable coherent tetragonal-ZrO $_2$ precipitates having diameters of \sim 250 nm and aspect ratios of \sim 5:1. The lenticular t- ZrO $_2$ precipitates occupy 40–50 vol.% of the c-ZrO $_2$ grains [70]. Because Mg-PSZ is transformation-toughened by these

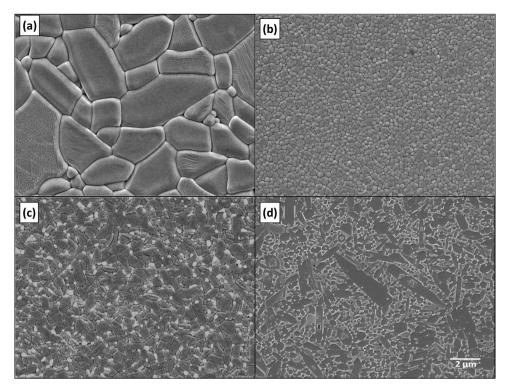


Fig. 1. Microstructures of biomedical ceramics: (a) Al₂O₃; (b) Y-TZP; (c) ZTA; and, (d) β-Si₃N₄; Courtesy Kyocera and Amedica Corporations.

nano-sized acicular t-ZrO2 precipitates, grain size control of the larger c-ZrO₂ is less important. Trials were conducted in the 1990s in an effort to decrease the grain size of Mg-PSZ by adding a small amount of yttria and/or a second phase (alumina or an aluminate) [201,202]. The purpose of these additives was to limit grain boundary mobility. However, follow-up studies were limited because Y-TZP was considered the system of choice during that period. Because of concerns with LTD in Y-TZP, there might be renewed interest in these Mg-PSZ systems, with the goal of improving their strength by careful control of their microstructure.

Grain refinement is essential for Y-TZP, ZTA, AMC, and ATZ composites, but for one additional key reason. The transformation of the t-ZrO₂ phase within these materials occurs more readily when ZrO₂ grains are beyond a critical size [203]. This size for Y-TZP under hydrothermal conditions is reportedly less than 360 nm [204]. Deville et al. confirmed that small ZrO₂ grains < 500 nm are also required in ZTA ceramics [205], with the Al₂O₃ matrix influencing their critical transformation size. Typically, the t- ZrO2 grains are located at alumina grain boundaries or at triple-point grain junctions [206]; but unique intragranular composites have been developed having submicron Al₂O₃/t-ZrO₂ grains embedded within larger Al₂O₃ grains [207]. The concentration and dispersion of the t-ZrO₂ grains within the Al_2O_3 matrix (or the converse for ATZ composites) is important for optimum strength. Adequate intermixing of the Al₂O₃ and ZrO₂ ceramic powders prior to densification is essential [208]. For ZTA composites, where Al₂O₃ is the dominant phase, keeping the ZrO₂ concentration below the theoretical percolation limit (~16 vol.%) prevents linking-up of t-ZrO₂ grains throughout the Al₂O₃ matrix [205]. Their isolation is important in inhibiting intergranular H₂O diffusion which leads to premature transformation. AMC composites have an additional tertiary phase engineered into their microstructure composed of magnetoplumbite platelet grains [209]. Their purpose is to increase the ceramic's overall resistance to fracture via crack deflection [152,156], which is plausible in principle, but their actual contribution to increased toughness, considering their small volume percentage in the composition, has

not been fully verified [210]. For ATZ, where t-ZrO $_2$ is the matrix and Al $_2$ O $_3$ is the minor phase, strength improvements are due to transformation toughening and intergranular crack deflection induced by the presence of the Al $_2$ O $_3$ particulates at t- ZrO $_2$ grain boundaries [211,212]. Dispersion of the Al $_2$ O $_3$ and ZrO $_2$ grains is therefore an important design element of this material. However, this ceramic is also highly susceptible to hydrothermally induced phase transformation because of the large proportion of t-ZrO $_2$ grains, although its kinetics are reportedly retarded when compared to Y-TZP ceramics [162]. Representative microstructures of Y-TZP [213] and ZTA [154] are shown in Fig. 1(b) and (c), respectively.

Grain size control is also important for Si₃N₄, yet it is not as critical as it is for transformation-toughened materials. Because β-Si₃N₄ grains are acicular, it has been shown that bimodal structures [214] or large aspect ratio grains [120-122] are beneficial in improving fracture toughness through a number of mechanisms, including crack deflection, crack bridging, grain pull-out, and compressive stresses [122]. The acicular grain structure of Si₃N₄ is unique among bioceramics. Elongated grains are generated during densification via a dissolution re-precipitation process that irreversibly converts equiaxed α-Si₃N₄ particles to their acicular β -Si₃N₄ counterparts. The β -Si₃N₄ grains impart high toughness and strength comparable to transformation-toughened ceramics without the inclusion of a metastable phase. Consequently, Si₃N₄ is not susceptible to hydrothermal degradation. A representative microstructure of a biomedical Si₃N₄ ceramic is provided in Fig. 1(d).

A summary of grain sizes for the various ceramic coatings is provided in Table 1b. PVD ceramic coatings are either amorphous or nano-crystalline. Their stoichiometry can vary depending on chemistry and deposition conditions [215]. For instance, DLC coatings are amorphous but can differ in their atomic bonding structure. Codeposition of carbon with hydrogen results in softer sp² graphitic type bonds, whereas pure carbon deposition can produce hard sp³ bonding with corresponding hardness values ranging from about 14 GPa to 80 GPa [148,216]. Other PVD coatings such as TiN,

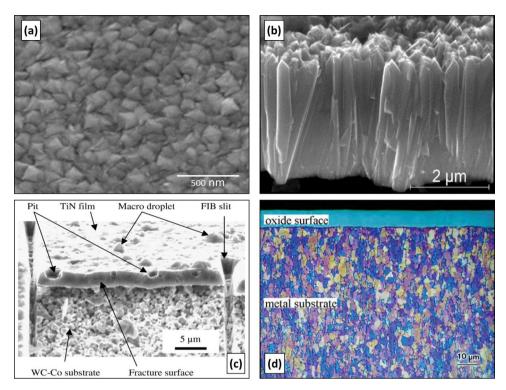


Fig. 2. Microstructures of ceramic coatings on metallic substrates: (a) TiN coating at 1.26 μm thickness [226]; (b) cross-section of TiN coating [227]; (c) fracture surface of TiN coating showing the presence of pit and droplet defects [228]; and, (d) cross-section of Ox–Zr *in situ* grown layer [180]. Reprinted with permission.

ZrN, and TiNbN possess nano-crystalline columnar microstructures [129,208,209]. Their acicular size can be 30×300 nm, with the thickness of the grains increasing proportionately in the crystallite growth direction. It is also common practice within industry to deposit multiple transitional layers to increase substrate adhesion [217–219]. For example, in coating a CoCr alloy with ZrN, one manufacturer includes a bond layer and up to five interlayers, alternating between CrN and CrCN, with the final surface layer being ZrN [134,220]. The initial layer, which is generally a compositional variant of the substrate, is designed to achieve good chemical bonding while subsequent layers modulate a transition in elastic properties. These layers reduce deposition and thermally induced compressive residual stresses, which can be significant—up to 11 GPa [221]. Delamination of the coating would readily occur in their absence [222]. The composition of these layers is experimentally determined, with the thickness of the total deposition typically being less than about 15 mm. Most PVD coatings also contain undesirable micro-droplets as a result of the deposition process. These are larger particles ablated from the PVD target and accelerated toward the substrate within the plasma. They result in degradation of the smoothness, uniformity and thickness of the deposited layer, and can be a source of mechanical or corrosive failure [223,224]. Elimination of 100% of these defects is statistically improbable, and special filters have been added to PVD equipment to minimize their occurrence [225]. Examples of the surface texture and crosssection of two different TiN coatings are shown in Fig. 2(a) and (b), respectively [226,227], and a graphic example of coating defects is shown in Fig. 2(c) from the work of Kamiya et al. [228].

A detailed examination of the grain morphology of OxZr was performed by Hobbs et al. [190] This *in situ* grown layer predominately consists of nano- crystalline columnar monoclinic $\rm ZrO_2$ grains that are about 20–70 nm wide \times 200–300 nm long, arranged in an anisotropic brickwork pattern, with their long axis orthogonal to the surface of the Zr–2.5Nb substrate. Fortuitously, this aligned grain structure resists crack propagation parallel to the alloy's surface, thus reducing the layer's chances for delamination.

Also, their large aspect ratio and parallel alignment advantageously resist pull-out from the polished articulation surface. Adhesion of the oxidized layer to the substrate is facilitated by Nb stringers which extend from the metal into the oxide and serve as anchors. Compressive oxidation surface stresses, reportedly $\sim\!670\,\mathrm{MPa}$, further assure coherency of the monoclinic ZrO₂ layer [190]. A typical microstructure of OxZr at the Zr–2.5Nb alloy interface is presented in Fig. 2 (d) [180].

The grain structure of hydroxyapatite coatings is lamellar due to the nature of the deposition process. Plasma spraying results in partial or full melting of the hydroxyapatite particles. Demnati et al. suggest that the compositional structure of the particles is composed of concentric layers of different apatite compounds as shown in Fig. 3 [63]. The outermost layer is CaO and a melt, followed by successive inner layers of tri- and tetra-calcium phosphate (TCP, TTCP) and orthohydroxyapatite (OHAp). At the core of the particle is solid hydroxyapatite (HAp). Upon impaction, partially melted particles solidify as multiphase mixtures in small-sized deformed splats at the substrate's surface. The size of these splats is proportional to the original HAp particles and can range up to about 100 µm. Successive particles impacting on their predecessors form the lamellar structure [229]. The initial lamellae are different from their successors since the first layer is deposited on the high thermal conductivity metallic surface, whereas subsequent layers adhere to their thermally insulating predecessors. The apatite composition of the coating changes with successive deposition passes as indicated in Fig. 3.

Differential cooling and reheating is responsible for migration and segregation of the apatite phases within the coating. The amorphous phase (ACP) dominates the composition in the as-sprayed condition; but controlled crystallization of ACP can be accomplished by post deposition thermal treatments [230]. Porosity is generated within the coating during cooling due to entrapped air, and cracks form due to differential shrinkage. Microstructures for a thermally sprayed HAp coating are shown in Fig. 4(a) and (b) [229]. Newer HAp coating compositions incorporate ZrO₂ or

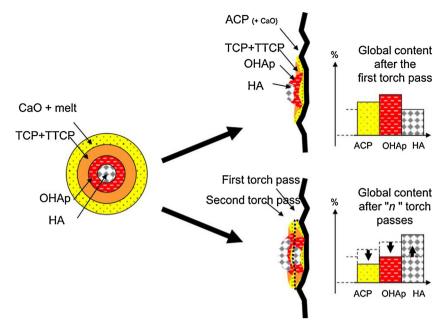


Fig. 3. Sequence of events during deposition of a semi-molten HAp particle impinging on a roughened surface [63]. Reprinted with permission.

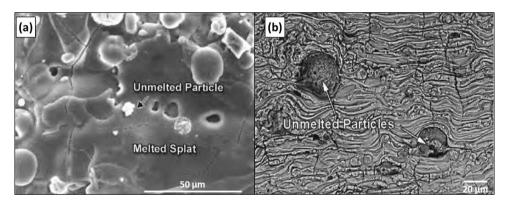


Fig. 4. Flame-sprayed HAp coating: (a) surface grain structure; (b) cross-section of grain structure [229]. Reprinted with permission.

Ti6Al4 V as additives to improve substrate bonding and increase coating toughness [231]. More recently, HAp has also been coated onto, or incorporated into polyetheretherketone (PEEK) to improve osseointegration of spinal fusion cages. In the former case, HAp coatings are applied using PVD methods [232,233]. In the latter instance, HAp particles are compounded into PEEK prior to thermal forming and machining of the implant [234,235,236].

In summary, microstructural engineering has been essential in enabling and improving biologic functions of ceramics and ceramic coatings. Reduction of grain size is necessary for increased strength and toughness in Al₂O₃. Its control is also critical for phase transformation toughening at the crack tip under applied stresses and the prevention of premature transformation under hydrothermal conditions in Y-TZP, ZTA, AMC, and ATZ compositions. Conversely, growth of acicular β- Si₃N₄ grains to an appropriate aspect ratio results in increased strength and toughness without the necessity of a metastable transformable phase. Microstructures for PVD coatings are amorphous or nano- crystalline, and vary in size based on chemistry and deposition conditions, but micro-droplet defects and high residual deposition stresses are common. Improved substrate adhesion via incorporation of multiple interlayers is employed to reduce stress and modulate differential elastic properties. OxZr microstructures are also nano-crystalline with excellent substrate adhesion and anisotropic grain orientation that resists delamination and grain pull-out. Plasma sprayed HAp (or HAp mixtures with toughening additives) coupled with post-deposition heat treatments provide microstructures possessing adequate bond strength and fortuitous mixtures of apatite phases, porosity, and other defects for effective osseointegration. Finally, HAp has been incorporated into PEEK devices, either as a PVD coating, or as particles compounded with the polymer.

3.3. Strength

Strength is the ability of a material to resist deformation or fracture. The common method for measuring bulk strength of ceramics is flexural testing, which involves symmetrical bending of carefully prepared parallelepiped specimens in accordance with international standards [237,238]. Shown in Table 1a are compiled flexural strengths for ceramics, which range from 400 MPa for Al $_2$ O $_3$ to 1500 MPa for Ce- or Y-TZP. While this is indeed a broad range, differences in reported strengths of up to 90% have also been observed for some individual ceramics, particularly Mg-PSZ, Ce-, and Y-TZP. This spread in data is not only due to variations in material and processing, but also a result of differences in testing methods. Three specimen geometries and the use of either 3- or 4-point bending are allowed within the standards. Due to the stochastic nature of brittle fracture, maximum strengths are always

observed when using the smallest specimens in 3-point bending. Consequently, some of the highest reported strengths in Table 1a were obtained in this manner. Using $\mathrm{Si}_3\mathrm{N}_4$ as a model material, Katayama et al. found up to a $\sim\!60\%$ difference in strength between 3- and 4-point flexural testing, when using the smallest and largest allowable specimens, respectively [239]. However, they and Quinn et al. [240] pointed out that reconciliation of differing data can be accomplished through a simple relationship, provided an adequate number of specimens have been tested and a meaningful Weibull modulus determined. Nevertheless, this reconciliation might be more difficult in highly transformable ceramics, such as Ce-TZP, where the maximum failure stress relates to the transformation itself rather than to pre-existing defects.

 Al_2O_3 has a modest flexural strength of about 580 MPa, (cf., Table 1a) [40]. The ASTM and ISO specifications for this material are listed as \geq 400 MPa and \geq 500 MPa, respectively (cf., Table 2). Al_2O_3 's lower strength is the primary reason it has lost favor in recent years for use as femoral heads and liners, having been replaced by the higher strength and toughness ZTA and AMC ceramics. Nevertheless, in vivo fractures of Al_2O_3 components are now extremely rare. Reported failures declined from about 13.4% for implants produced prior to 1990 to 0.021% for parts produced between 2000 and 2013 [40,241]. This reduction was due to the aforementioned process improvements; and suggests that even a modest strength material can provide excellent in vivo reliability.

With the advent of Mg-PSZ and Y-TZP, flexural strengths improved markedly—from 700 up to 1500 MPa, respectively (*cf.*, Table 1a) [65,70,187]. However, their ASTM specifications were only set at 600 and 800 MPa, respectively (*cf.*, Table 2). Y-TZP might have been one of the best materials for articulation, were it not for its susceptibility to LTD. An incompletely validated process change by the leading manufacturer of Y-TZP in the late 1990s led to a dramatic increase of *in vivo* fractures, from 0.002% prior to the process change to 8% afterwards [73], resulting in their recall from US and EU markets in 2001 [242].

The introduction of ZTA and AMC composites was a significant step forward in joint arthroplasty due to a combination of improved flexural strength and reduced sensitivity to hydrothermal degradation. ZTA strengths as high as 1248 MPa have been observed, and flexural strengths of up to 1400 MPa have been reported for AMC on a series of \sim 40 production lots (cf., Table 1a) [154,243]. In vivo fracture rates for AMC are extremely low, at 0.001% [40]. It is now the preferred ceramic worldwide for hip joint replacements, although its LTD resistance remains an area of long-term research [79,172,244–248]. In spite of their high reported values, the minimum flexural strength for ZTA and AMC ceramics per the ISO standard was set at \geq 750 MPa (cf., Table 2).

Only limited strength data are available for ATZ composites, with values ranging between 755 and 1163 MPa (*cf.*, Table 1a) [249,250]. The lower average value was determined from 4-point bending of standard test bars, whereas the higher strength was obtained by biaxial flexural testing [251]. One manufacturer lists their ATZ's biaxial strength at 2000 MPa; but this extraordinary value is not found in the technical literature [252]. ATZ orthopaedic devices clearly show a certain degree of LTD since their major phase is Y-TZP [81], although the transformation kinetics appear to be slower than Y-TZP and the impact less dramatic.

 Si_3N_4 is the latest entrant in load-bearing medical applications. It is a non-oxide ceramic with flexural strengths comparable to the oxide materials, typically between 800 and 1100 MPa (cf., Table 1a) [82], with ASTM or ISO specifications of \geq 760 MPa (cf., Table 2). It has an advantage over ZTA and AMC ceramics in that its high strength is not reliant on the presence of a metastable transformable phase. It is therefore immune to LTD. Its acicular microstructure is responsible for its high flexural strength. An advancing crack must follow a tortuous and chaotic high-energy

path in its propagation. Grain bridging and pull-out in the crack wake serve to reduce tensile stresses at the crack-tip resulting in increasing resistance to crack propagation during crack extension, which is known as *R*-curve behavior [253].

Strength testing of PVD ceramic coatings is problematic due to their thin nature, small specimen sizes, substrate adherence, and existence of residual internal stresses. As reviewed by Borrero-Lopez et al., a number of techniques have been explored, including tensile, flexure, nano-indentation, and scratch methods [254]. They reported characteristic strengths for TiN and DLC coatings ranging from 1.25 GPa to 7.6 GPa. Jaeger et al. developed a novel flexural technique using pre-cracked metallic substrates and measured strengths for TiN coatings from 183 to 491 MPa. They concluded that their results were reasonable based on comparable data for HIPed TiN. Separately, Wiklund et al. used four-point bending inside an SEM to evaluate fracture resistance of TiN thin films using relatively large specimens (i.e., 60 mm²), and measured flexural strengths of between 600 MPa and 1.0 GPa [255]. Later, Kamiya et al. prepared free-standing TiN films and performed flexural tests with strengths varying from 2.60 to 3.87 GPa. However, the area of their specimens was only 1 µm². They rationalized that their higher strengths were not inconsistent with Wiklund's findings after accounting for surface area differences [228]. Finally, for DLC films, Espinosa et al. used nano-indentation to measure fracture strength and obtained values of 4.7-5.2 GPa on samples with surface areas ranging from $500 \, \mu m^2$ to $4000 \, \mu m^2$ [256].

This broad and inconsistent range of observed strengths highlights the difficulty in obtaining quality data for coatings. Consequently, scratch adhesion testing was developed as a practical engineering method for assessing coating viability [257]. This technique involves dragging a diamond stylus across the coating's surface under a linearly increasing load. Typically, two critical loads $(L_{C1} \text{ and } L_{C2})$ are determined by examining the scratch track for brittle damage events such as cracking, delamination, chipping, spallation, or buckling. L_{C1} is recorded at the first sign of cohesive failure, with L_{C2} noted at complete coating failure. One or both of these values are often reported in the literature, with L_{C2} typically designated as the coating's scratch adhesion strength. It is important to note that this test does not provide intrinsic scientific information on adhesion. Rather, it yields valuable comparative engineering data. As pointed out in the standard, test outcomes are dependent upon a number of factors including stylus properties, geometry, loading and displacement rates [257]. With this background information, shown in Table 1b is the range of scratch adhesion values reported in the literature for various hard ceramic coatings. Values from 10 N to 160 N were observed. Differences are due to deposition chemistry and methods, coating thicknesses, and substrate materials.

Hardness of the substrate plays a major role in determining scratch adhesion. As demonstrated by Roy et al., hard substrates perform better than soft ones. Their study correlated scratch adhesion data for DLC coatings applied to Mg-PSZ and CoCr substrates, with hardness values of 10-12 GPa and 3-4 GPa, respectively. Also included was an OxZr coating with a Zr-2.5Nb substrate hardness of 2.5 GPa. L_{C2} values of 46.8 N were obtained when the DLC was applied to Mg-PSZ versus 35.2 N on CoCr, which was remarkably close to the OxZr result of 34.8 N [258]. A similar conclusion was reached by Utsumi et al. after applying an interlayer of hard tungsten carbide between an aluminum substrate and a thin DLC coating. Scratch adhesion values were increased from about 12 N to 80 N [259]. These results are not surprising given that a hard brittle ceramic coating on a soft metallic substrate is analogous to applying a "hard candy coating over soft chocolate." Under an applied load, slight elastic deformation will readily fracture the thin brittle coating. Even though scratch adhesion does not determine the intrinsic strength of thin films, it is an important engineering measurement in predicting *in vivo* viability. As will be discussed in more detail later in this review, one of the failure modes for coated orthopaedic devices is poor adhesion. Adhesion can be significantly improved by closely matching elastic properties between the coating and substrate, and by ensuring strong chemical and mechanical interfacial bonding.

Strength testing of HAp coatings is also problematic due to their thin nature and inherently defective structure. However, flexural tests have been obtained for bulk HAp densified using conventional sintering, spark-plasma sintering, and hot-isostatic pressing with reported values ranging from 113 MPa to 189 MPa [260-262]. Strengths were increased up to 220 MPa by reactive sintering of HAp and fluorapatite admixed with up to 40 vol.% ZrO₂ [263], and values of ~250 MPa were obtained when using 40 vol.% Al₂O₃ [264]. Even higher strengths of about 300 MPa were obtained when combining HAp with 30 vol.% Al₂O₃ and 15 vol.% ZrO₂ [265]. While these compositional studies on bulk HAp are instructive, they are not directly transferrable to coatings, and strength testing of coatings is essentially impossible. Scratch adhesion can be employed, but the more practical approach for evaluating HAp coatings is to utilize "pull-off" testing in accordance with international standards [266,267]. Specimen preparation and testing is straightforward. After applying the HAp coating to one end of a 25.4 mm diameter metallic rod, an identical rod is bonded to the coating using adhesive glue. The joined test specimen is then subjected to tensile loading, and the "pull-off" load recorded. The test yields valuable engineering data about the bond strength of the coating; but, similar to scratch adhesion, it does not provide scientific data on the strength of the coating itself. The standards require that the "pulloff" adhesion strength be greater than 15 MPa, which is one-fourth to one-tenth of the reported flexural strengths for dense HAp. Different deposition practices have been explored to improve "pulloff" strengths, including plasma, induction, vacuum-aerosol and high velocity oxy-fuel combustion spraying, hot-isostatic pressing, dip coating, electrophoretic deposition, sol-gel adhesion, pulsedlaser, ion-beam, or co-spraying HAp with a toughening agent such as Al₂O₃, ZrO₂, or a titanium alloy [268–272]. However, none of these processes produce "pull-off" strengths approaching flexural data for bulk HAp, with values ranging between about 10 MPa and 80 MPa [54,268]. The highest reported "pull-off" strengths of 60-80 MPa were produced using either a high power laminar jet coating method or vacuum deposition [268,272]. Shown in Fig. 5 is a range of "pull- off" adhesion strengths obtained using various deposition techniques [268]. It is easy to understand why both bulk HAp and HAp coatings cannot be used as structural members in orthopaedic devices given these observed low flexural and "pull-off" strengths.

This section on strength was initiated with a discussion of observed fracture rates for ceramics in THA. Even though their occurrence is small, they are serious events, requiring immediate revision surgery, debridement of damaged tissue, and replacement of THA components. Nevertheless, it is now evident that failure rates for metals have eclipsed those of ceramics. National registries and follow-up studies indicate that the hip systems with metal-on-metal (MoM) articulation have higher revision rates than any current ceramic based devices [273-275]. Fig. 6 shows survivability statistics for a combination of cemented and uncemented THAs from the UK national joint registry [273]. The revision rate for MoM articulation is over four times that of ceramic containing devices at ten-years postoperatively. The pathology of these MoM failures is not catastrophic fracture, but rather, excessive component wear, which symptomatically presents itself as groin pain, component loosening, metallosis, formation of pseudotumors, and localized tissue necrosis, with elevated serum levels of Co and Cr ions. Patient morbidity and the requirement for revision surgery to replace worn metal components are no less serious, and in many

instances more severe than for ceramic fractures. In retrospect, the harm and suffering endured by patients with MoM devices might have been substantially eliminated by using ceramic-based THA systems. Keep in mind that the strength of ceramics is even higher than most metals. The issue is their low strain to failure as a consequence of high elastic modulus and lack of plasticity. Material strength is therefore only one of many factors that should be considered in the selection of appropriate orthopaedic implants.

3.4. Weibull modulus

A biomaterial's Weibull modulus provides a statistical measure of its strength variability. The range of observed strengths for a brittle material is related to the number and size distribution of its inherent flaws. As reviewed by Quinn et al., [240] the two parameter Weibull distribution function is commonly used to characterize experimentally determined strengths. The probability of a failure, $P_{\rm f}$, under an applied stress, σ , is given by the following expression:

$$P_{\rm f} = 1 - \exp\left(-\int_{\nu} \left(\frac{\sigma}{\sigma_0}\right)^m \mathrm{d}V\right) \tag{1}$$

where σ_0 is the Weibull scale parameter or characteristic strength, which is a measure of the centrality of the distribution and represents a failure probability of 63%. The Weibull modulus, m, is an indicator of the dispersion in strength values. A material with a high modulus has a narrower strength distribution and improved reliability. V is the volume of the specimen or component. The two parameters, σ_0 and m, are determined from strength testing followed by graphical interpretation of the results in accordance with an international standard [276].

A wide range of Weibull moduli is reported for bioceramics (cf., Table 1a). Values between about 7 and 18 are nominally observed, but measurements from 5 to 87 have been reported. Most ceramics fall within the nominal range, and repetitive measurements on serially produced batches and round-robin studies confirm this fact [239,277,278]. A low value of 5 was reported for Al_2O_3 [171], but this appears inconsistent when compared with the aforementioned repetitive studies. Extreme values of 53 and 87 were found for Si₃N₄ and ZTA, respectively, but these are not routinely observed for standard production batches [279,280]. Note that the ASTM and ISO specifications for Al₂O₃, Mg-PSZ, Y-TZP, ZTA, and AMC have been set at $\geq 8-10$ (cf., Table 2). Conversely, for Si₃N₄, its Weibull modulus was set at ≥ 12 , (cf., Table 2). Si₃N₄'s elongated grain structure can influence its Weibull modulus. Increases in size and aspect ratio of grains result in higher fracture toughness and strength due to R-curve behavior [281–283]. Without acicular growth, the mechanical properties of equiaxed Si₃N₄ are similar to oxide ceramics [284,285]. Also, Weibull moduli concomitantly increase for larger high aspect ratio grains because they become strength limiting flaws [283].

There is a paucity of Weibull strength data for ceramic coatings due to the difficulty in preparing specimens and conducting tests (*cf.*, Table 1b). This is particularly true for PVD thin films. For instance, Borreo-Lopez et al. assessed a number of hard ceramic coatings, including DLC and TiN and found Weibull moduli of 6–12 and 10–18, respectively [254,286]. Their results are consistent with Kamiya et al., who conducted flexural fracture tests on TiN films and found their moduli varied between 5 and 12 [228]. No Weibull data were found in the literature for ZrN, TiNbN, or OxZr thin films. Limited Weibull statistics have been compiled for HAp using biaxial flexural strength (BFS) and nano-indentation hardness measurements. Weibull moduli for strontium doped HAp fell within the range of 4–19 from BFS tests [287]. Using nano-indentation, Yang et al. found Weibull moduli ranging from 3 to 4 [288], and Dey et al. measured values between 2 and 9 [289]. Overall, the range of

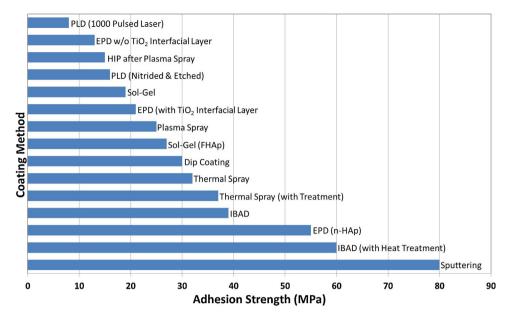


Fig. 5. HAp adhesion strengths for different coating methods. Adapted from Mohseni et al. [268] Used with permission.

results for coatings is not markedly different than those observed for bulk ceramics.

Consistently attaining both high characteristic strengths and Weibull moduli on sequential production ceramic batches is a significant challenge for process engineers. Strength-limiting defects can occur at any manufacturing step, but are most likely to be entrained during powder processing (including spray- drying). Coarse particles (or granules), large or hard agglomerates, and foreign contamination are common flaws. Inadequate particle packing after compaction further exacerbates their negative contribution, leading to formation of pores or voids during sintering [290].

Hot-isostatic pressing was largely implemented to minimize or eliminate void-related defects [187]. Surface cracks and pull-outs can be generated during post-densification grinding operations [291–293].

Furthermore, specimens used in assessing mechanical properties have different processing than components. Therefore, their characteristic strengths and Weibull moduli may not be representative of actual implants. This was the case for the recalled Y-TZP femoral heads in 2001. The defects resulting in their failure were on uninspected ID surfaces [73]. For coatings, inappropriate deposition conditions can generate droplets, or produce voids, inclusions,

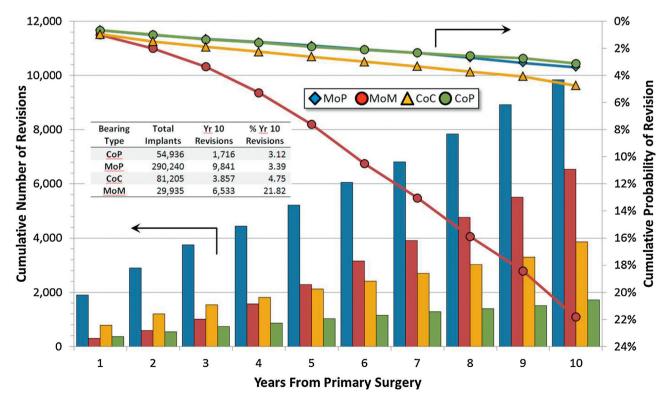


Fig. 6. Cumulative number and percentage of revisions for cemented and uncemented hips with different bearing surfaces (MoP = metal-on-polyethylene, MoM = metal-on-metal, CoC = ceramic-on-ceramic, and CoP = ceramic-on-polyethylene) [273].

and pits, all of which reduce reliability. Low characteristic strengths and Weibull moduli are signals to process engineers that special cause variation has occurred, requiring investigation and corrective action.

Alternatively, an increase in characteristic strength and Weibull modulus can be accomplished by truncating the flaw population through the use of proof-testing [198,294,295]. Proof-testing is an inspection technique that subjects finished components to a stress in excess of physiologic loads. Proof-testing eliminates a priori any component that has unacceptable strength, and therefore provides a threshold stress, below which no failure is likely to occur. Today, ceramic THA components and spinal fusion cages are subjected to proof-testing. Proof-test levels are based on finite element analyses, or in vitro and in vivo studies, and regulatory requirements. Because of this fact, the measurement of a Weibull modulus, while important for process investigation and control, has less meaning and is less relevant as a design parameter. Consequently, as shown in Tables 1a and 1b, the reliability of ceramics and coatings, as expressed by their respective Weibull moduli, is essentially equivalent.

3.5. Fracture toughness

Fracture toughness is a fundamental property governing strength. It describes the material's intrinsic resistance to crack propagation. While it is applicable for all materials, it is critically important for ceramics. The strength of a ceramic is related to its fracture toughness by the Griffith equation [296]:

$$\sigma_{R} = \frac{K_{IC}}{Y\sqrt{\pi a}} \tag{2}$$

where σ_R = fracture strength; K_{IC} = fracture toughness, or critical stress intensity for fast fracture; Y = a geometrical factor related to the crack; and, a = crack length. Fast fracture occurs when the stress intensity at the crack tip exceeds the material's innate fracture toughness, K_{IC} . From the Griffith equation, there are two routes for improvement of a ceramic's fracture strength. One method is to control processing of the material in order to reduce pre- existing defects. This approach was discussed in the prior section on Weibull modulus. The second method is to engineer the microstructure to increase its K_{IC} .

A myriad of procedures for determining fracture toughness of ceramics have been developed. Common methods include chevronnotch (CVN), indentation fracture (IF), indentation-strength (IS), single-edge notched beam (SENB), single-edge v-notched beam (SEVNB), single-edge pre-cracked beam (SEPB), surface- crack in flexural (SCF), and Vickers indentation (VIF), with newer techniques being developed periodically - the details of which are beyond the scope of this review. The reader is referred to critical commentary by a number of noted fracture mechanics authorities for in-depth discussions of these various methods [297-301], including their applicability to biomaterials [302]. Round-robin testing was conducted in the 1990s to evaluate, reconcile, and standardize fracture toughness procedures [303-306]. Shown in Fig. 7 are results for five different ceramic materials - sintered silicon carbide (Si-SiC), alumina (Al₂O₃), Y-TZP, hot-pressed silicon nitride (HPSN), and magnesia partially stabilized zirconia (Mg-PSZ), tested in ten laboratories using five methods – CVN, IF, IS, SEPB, and SENB [304]. Note that differences between techniques can range up to 95% for some materials. Conclusions from this, and the other cited studies, highlighted inadequacies with indentation methods (IF, IS and VIF), and preferences for CVN, SEPB, and SENB techniques which were judged to produce the most consistent results. These three techniques were eventually incorporated into an ASTM standard with recommendations to utilize at least two methods for assessing toughness of unfamiliar or new materials [306].

A number of strategies have been devised to enhance toughness in ceramics. For single phase materials with equiaxed structures, grain size control and grain boundary engineering are performed, with sub-critical-micro-cracking, crack-bridging, and branching yielding toughness improvements [307,308]. Transformation toughening is used in Mg-PSZ, Y- and Ce-TZP, ZTA, AMC, and ATZ; whereas in situ growth of elongated grain structures in Si_3N_4 increases its K_{IC} . Composites incorporating metal or ceramic dispersoids, platelets, whiskers, fibers, or laminates also lead to increased toughness by crack deflection and branching [307]. Fracture toughness values for the ceramics are listed in Table 1a. Al₂O₃ has a fracture toughness between 3.3 and 4.2 MPa $m^{1/2}$, which is the lowest among the structural ceramics used in joint arthroplasty [309,310]. It also has the lowest ISO specification at 2.5 MPa $m^{1/2}$ (cf., Table 2). Its toughness is related to its microstructure, which consists of equiaxed grains of less than about 2 µm. Strength improvements in Al₂O₃ have not been due to marked increases in fracture toughness [206]. Rather, they have been achieved by controlling processing defects and decreasing grain size. As reviewed by Pezzotti et al., thermal expansion anisotropy within the corundum crystal structure between the c-axis and aor *m*-axes leads to weak grain interfaces in the densified ceramic [311]. These interfaces become defect origins under applied tensile loads, with crack propagation typically occurring along grain boundaries. Consequently, larger grains (usually referred to as "abnormally grown" ones) not only originate fractures, but also allow a crack to easily propagate along very weak or even already broken grain boundaries. With a smaller grained material, this detrimental phenomenon is less effective; but the crack possesses a less tortuous path through the matrix. However, it is principally a difficulty in engineering the strength of grain boundaries while concurrently controlling the grain morphology that limits the toughness of Al₂O₃. On the other hand, Mg-PSZ, Ce- and Y-TZP, ZTA, AMC, and ATZ ceramics also have equiaxed grains, but the polymorphic expansion of the ZrO_2 crystal lattice during the $t \rightarrow m$ transformation shields the crack-tip from excessive tensile forces, thereby slowing or arresting its progression. Because of this, K_{IC} values for these ceramics are higher than for Al₂O₃, ranging from 4.1 up to 20.0 MPa $m^{1/2}$ [153,312]. The ASTM standards are silent with respect to a fracture toughness specification for Mg-PSZ and Ce- or Y- TZP, whereas the ISO standard for ZTA and AMC materials was set at 3.5 MPa $m^{1/2}$. Si_3N_4 has the highest ASTM or ISO specification for fracture toughness, at ≥ 6 MPa m^{1/2} (cf., Table 2), with reported K_{IC} values between 4.4 and 15.0 MPa m^{1/2} [304,313]. This range is not only due to testing differences, but also from its diverse microstructures, which, depending on composition and processing, can be engineered to have smaller more equiaxed or larger high aspect ratio grains. Increased toughness from the latter type of microstructure is due to its interlocking nature. Crack tortuosity and bridging grains in the crack wake reduce critical stress intensities at crack tips by absorbing significant tensile energy (up to 1.5 GPa), which enhances R-curve behavior [308,313]. Indeed, Si₃N₄ has the strongest R-curve of any ceramic, with K_{IC} values rising from 4 to 15 MPa m^{1/2} over crack extensions of \sim 800 μ m [313]. *R*-curve trends for Al₂O₃, Y-TZP, ZTA, and two Si₃N₄ materials are given in Fig. 8 [311].

Acquisition of fracture toughness data for ceramic coatings suffers from the same difficulties encountered in assessing their strength. Measuring $K_{\rm IC}$ values is challenging given their thinness and bonding to softer metallic substrates. Techniques including bending, buckling, scratching, and indentation have been used, but as of yet there are no standards nor consensual methods [314,315]. Scratch adhesion and Vickers indentation are most practiced because of their operational simplicity. Empirical formulas are employed for scratch adhesion, which arguably are more a measure of load carrying capability than intrinsic toughness. Indentation

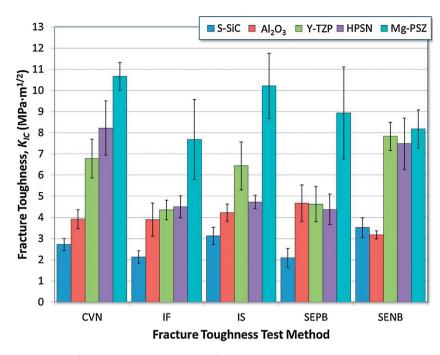


Fig. 7. Results from round robin evaluation of different methods of assessing fracture toughness [304].

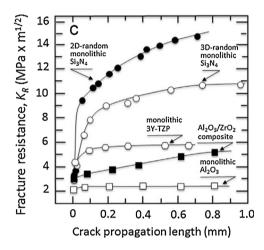


Fig. 8. Comparative *R*-curve measurements of bioceramics [311]. Reprinted with permission.

methods which use the equations and measurement methods for ceramics suffer from the same inconsistencies [316]. Due to their thinness, nano-indentation is often employed to avoid substrate interactions, with concomitant difficulties arising in the formation and measurement of acceptable crack lengths. Compounding these problems are issues associated with intrinsic coating defects and residual stress which also alter apparent $K_{\rm IC}$ values [315].

Increased fracture toughness is deemed to be important for coatings, particularly in tribology applications. As reviewed by Zhang et al., research continues to be conducted on improving toughness of these brittle films [222]. Not surprisingly, the envisioned techniques are similar to "tried and true" practices used for ceramics, including cohesive strengthening of grain boundaries and its juxtaposition of providing for grain boundary sliding, phase transformation, utilization of multiple layers, fiber reinforcement (*i.e.*, carbon nanotubes), compressive stress toughening, and incorporation of a ductile phase. The most practiced and therefore the most reported method for improving toughness is incorporation of multiple interlayers. Sequential layering of coatings with

varying hardness and metallic content was originally developed for the express purpose of increasing adhesion. However, it was soon realized that composite layered structures provide beneficial improvements in fracture toughness as well [222,317-319]. Provided in Table 1b are reported fracture toughness data for hard ceramic coatings, which range from 0.7 to 12.4 MPa m^{1/2} for TiN coatings [320,321]; and from 1.6 to 5.1 MPa m^{1/2} for DLC [316,322]; with ZrN between 2.3 and 7.5 MPa $m^{1/2}$ [323,324]. TiN is perhaps the most studied, and therefore it is not surprising that its observed $K_{\rm IC}$ range is broader than for any of the other PVD coatings. However, as pointed out by Zhang et al. [321] and by Mofidi et al. [325], toughness values for TiN films vary considerably based on film thickness and interlayers. There are no reported data for TiNbN, but its K_{IC} range will likely be similar to TiN. An investigation of the fracture toughness of OxZr layers was performed by Leto et al. using cathodoluminescence spectroscopy, resulting in K_{IC} values of between 2.2 and 2.8 MPa $m^{1/2}$ [326]. Their results are similar to values observed for bulk monoclinic zirconia [327].

To summarize, K_{IC} values for various biomaterials as a function of their elastic moduli are shown in the Ashby diagram of Fig. 9 [328]. This chart compares ceramics, metals, and polymers and contrasts them to cortical and cancellous bone. The chart highlights interesting engineering trade-offs in the selection of implant materials. Metals are an order of magnitude greater in toughness and elastic moduli than cortical bone. The structural bioceramics and ceramic coatings also have high moduli; but, with the exception of HAp, their toughness values are essentially equivalent to bone. Neither class of materials has a combination of elastic or fracture properties that closely match cortical bone. However, only engineered polymers have a combination of toughness and elastic moduli that approximate native bone; and yet these plastics are not exclusively used as structural replacements because of their low yield or fracture strengths. Consequently, joint arthroplasty systems have evolved to typically include a combination of two or more materials-polymers, ceramics, and metals.

To understand this evolution requires a brief review of the chemistry, structure, and properties of bone. Approximately 70 wt.% of bone is composed of a variant of HAp-nanocrystals, which are slightly deficient in calcium with added carbonate [55].

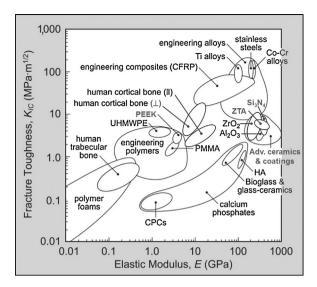


Fig. 9. An Ashby diagram showing the elastic modulus and fracture toughness of bone tissue compared to biomaterials commonly used as orthopaedic implants. The mechanical properties of cortical bone are shown for loading parallel (\parallel) and perpendicular (\perp) to the longitudinal anatomic axis. Adapted from Roeder et al. [328] Used with permission.

However, as previously noted, HAp is too brittle for use as a structural member alone. Consequently, the remaining 30% of bone is composed of collagen that is bound and supported by an arrangement of apatite nanocrystals, forming mineralized fibrils that are organized into lamellar structures. The basic metabolic unit is an osteon, which consists of vascular canals that are surrounded by the lamellae in concentric rings. Bone composition is essentially constant despite the fact that its properties vary based on location—from its dense and strong exterior cortical surface to its porous and flexible cancellous core.

Due to this architecture, its mechanical properties are generally reported as a function of density [329]. For cortical bone, alignment of the osteons with the bone's long axis bestows unique and beneficial anisotropy. Strength and elastic modulus are higher parallel to the axis [330–336]; whereas, hardness and fracture toughness are greater in the orthogonal direction [337] (cf., Table 1a). It is stiff and strong in its longitudinal direction and yet tough and flexible in its transverse direction. Although soft when compared to nonliving engineered materials, cortical bone of the femur is hardest along the diaphysis and the distal radius next to the articulation surface [338]. It exhibits remarkable R-curve behavior in its transverse direction, with toughness values increasing from approximately 5-20 MPa m^{1/2} under crack extensions of up to $400 \mu m$ [337]. It can withstand reasonable loads and deformation with no residual damage, and yet, absorb considerable energy through micro-cracking, crack deflection, fiber pullout, or crack bridging to prevent complete failure when stress levels are excessive [335]. Because it is living tissue, bone remodeling is a natural response to applied stress, allowing it to reshape and repair itself. Nature has indeed engineered bone to be an extraordinary responsive composite consisting of a native bioceramic (hydroxyapatite) and a native polymer (collagen fibrils). It uniquely combines appropriate mechanical attributes from each material into a living composite. It is therefore not surprising that modern arthroplasty designs typically consist of combinations of two or more engineered materials. Yet, because biomaterials are non-living, they lack the ability to self-repair or remodel, and therefore a level of additional safety is needed in their properties, particularly fracture toughness and strength, both of which must exceed native bone to ensure adequate performance and prevent premature failure.

3.6. Slow crack growth and fatigue resistance

It is well known that ceramics and glasses are susceptible to delayed failure at stress intensities below K_{IC} under fixed or dynamic loads. This failure phenomenon is often referred to as slow crack growth (SCG) or sub-critical crack growth. Slow crack propagation occurs through linking-up of inherent microstructural defects under static or cyclic stress. The reader is referred to the work of Ritchie et al. for a comprehensive review of the mechanics and mechanisms for fatigue in ceramics, with contrasts to similar processes in metals and composites [339–341]. In brief, there are two operative mechanisms in ceramics, designated as intrinsic and extrinsic, that either advance or retard crack propagation, respectively. Growth is promoted ahead of the crack tip by *intrinsic* factors such as processing defects (pores, cracks, inclusions, etc.) or due to weak grain boundaries, and is retarded in the wake of the crack by extrinsic factors that increase toughness (i.e., transformation- or microcrack-toughening, crack deflection, the presence of bridging grains, and frictional interfaces, etc.). As discussed in the prior section, K_{IC} is a basic material property, which determines a lower limit for instantaneous rupture. While SCG can be operative beneath this critical stress intensity, there also exists a stress intensity threshold below which no sub-critical crack growth occurs. This value is designated as K_{TH} . It is also a major basic material property and indicates the level of stress intensity that a material can withstand without SCG initiation (i.e., thus without delayed failure). Measurements of K_{TH} have traditionally been computed from cyclic stress rupture data, where the rate of crack propagation in a specimen is monitored as a function of load in accordance with the Paris equation [342]:

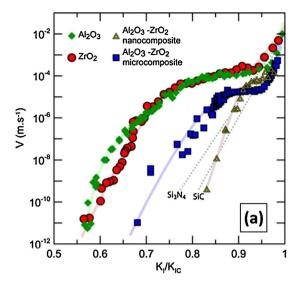
$$\frac{\mathrm{d}a}{\mathrm{d}N} = A(\Delta K)^m \tag{3}$$

where A and m are scaling factors specific to the material and test conditions, and da/dN is the crack growth rate, ΔK is the stress intensity range ($K_{\text{max}}-K_{\text{min}}$), and K_{max} and K_{min} are the maximum and minimum stress intensities applied during cyclic loading, respectively. In his review, Ritchie et al. further taught that K_{TH} was more sensitive to the maximum stress intensity, K_{max} , than to ΔK , and incorporated this into a modification of the Paris equation [339]:

$$\frac{\mathrm{d}a}{\mathrm{d}N} = C'(K_{\mathrm{max}})^n (\Delta K)^p \tag{4}$$

where C is a constant for the material and testing conditions, and (n+p)=m. For brittle materials, n is much larger than p, and therefore has a greater influence on K_{\max} . The SCG characteristics for a given ceramic can then be represented on a $V-K_{\rm I}$ diagram where the velocity of the crack, V or ${\rm d}a/{\rm d}N$ (on the y-axis), can be plotted against the applied stress intensity, $K_{\rm I}$ (on the x-axis). Traditionally, the lowest observed $K_{\rm I}$ value has been interpreted as the threshold stress intensity, $K_{\rm TH}$. While specific values of $K_{\rm TH}$ are often reported, a ratio ($K_{\rm TH}/K_{\rm IC}$) is usually also computed to normalize operative ranges for SCG. Doing so allows comparisons between different materials and testing methods. Using this methodology, available or computed ratios for $K_{\rm TH}/K_{\rm IC}$ based on graphical interpretations of $V-K_{\rm I}$ diagrams are provided in Tables 1a and 1b for ceramics and coatings.

For the materials of Table 1a, it can be observed that $K_{\rm TH}/K_{\rm IC}$ values lie between 0.37 and 0.97. This range is due to several factors. Results obviously vary by material, but can also differ due to test conditions and applied stress ratios. Furthermore, the data shown in the Table are inclusive of various environmental conditions – particularly the presence of moisture [211,212,343–355]. Stress corrosion cracking is a common problem for all polycrystalline ceramics due in part to the presence of grain boundary impurities.



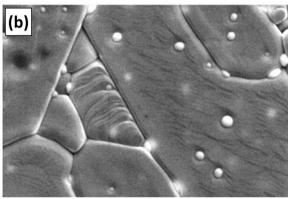


Fig. 10. (a) $V-K_1/K_{\rm IC}$ of biomedical grade alumina, yttria stabilized zirconia, conventional zirconia-toughened alumina 'micro'-composite, and nano-structured alumina-zirconia composite. Schematic $V-K_1/K_{\rm IC}$ of covalent ceramics (SiC and Si₃N₄) are given for comparison [310]; and (b) Microstructure of an Al₂O₃-ZrO₂ composite showing a nano-ZrO₂ grain imbedded in a larger Al₂O₃ grain. (a) Reprinted with permission; (b) courtesy INSA Lyon.

Ceramics with fewer impurities are less sensitive [212,345], and grain boundaries that have high nitrogen content show increased resistance to SCG [349,350]. Compositional and microstructural engineering of these advanced ceramics is on-going by many organizations in an effort to reduce their sensitivity to fatigue. For example, shown in Fig. 10(a) is a $V-K_{\rm I}$ diagram demonstrating \sim 50% improvement in K_{TH}/K_{IC} between a biomedical alumina and two ZTA composites, one of which has a highly refined microstructure [310]. Projected values for SiC and Si₃N₄ are also shown for reference purposes. This figure highlights intrinsic differences between oxide and non-oxide ceramics. Traditional oxide ceramics, such as Al₂O₃ and ZrO₂, have higher propensities for SCG than do non-oxides (i.e., SiC and Si₃N₄) because of differences in atomic bonding. Ionic bonds in oxide ceramics are particularly sensitive to attack by moisture at crack-tips (driven by a strong decrease in surface free energy), whereas covalent ceramics are more resistant to bond breakage [356]. Selection of covalent ceramics is therefore preferred for demanding structural applications. However, it is remarkable that the nano-Al₂O₃-ZrO₂ composite shown in Fig. 10(a) has SCG behavior similar to that of covalent ceramics [355]. Furthermore, this accomplishment is indicative of the importance of microstructural engineering and of its impact on mechanical properties. Shown in Fig. 10(b) is a representative microstructure of these types of Al₂O₃-ZrO₂ nano-composites. For reference purposes, fatigue testing has been

conducted for most of the biomaterials of Table 1a, and the reader is referred to selected publications for Al $_2$ O $_3$ [357–362], Mg-PSZ [363–365], Ce-TZP [366,367], Y-TZP [346,368–370], Ce- and Y-ZTA [212,348,355,371,372], ATZ [211], Si $_3$ N $_4$ [344,354,373,374], CoCr [375], Ti6Al4V [376,377], PEEK [378–380], and cortical bone [381–383].

Note from Table 1b that there is a conspicuous absence of $K_{\rm TH}/K_{\rm IC}$ results for coatings. Inherent fatigue data for coatings are difficult to obtain because of their thin nature. Indeed, excluding hydroxyapatite [384,385], fatigue results for coatings are more associated with the substrate, or the interface between the coating and the substrate, than the coating itself; and even here, there is a scarcity of reported values. The coating obviously imparts fatigue resistance to the substrate because of its combination of inertness, higher hardness, compressive stress, and adhesion [384,385]. However, if the coating delaminates *in vivo*, it can have a dramatic opposite effect, leading to poorer fatigue of the underlying metal [133]. Comprehensive reviews on coating/substrate fatigue, and corrosive fatigue of implantable metals have been provided by Sadananda et al. [386] and Antunes et al. [387].

Nowadays, a new method of assessing $K_{\rm TH}$ has been proposed that goes beyond the traditional graphical interpretations of $V-K_{\rm I}$ diagrams. It uniquely incorporates and demonstrates the importance of R-curves in determining fatigue resistance. First elucidated by Kruzic et al. [388] using Al_2O_3 and Si_3N_4 as model systems, and later expanded and verified by Gallops et al. [389], Hartelt et al. [390], and Greene et al. [391,392], it deconvolutes $K_{\rm TH}$ into two terms:

$$K_{\text{TH}} = K_{\text{IO}} + K_{\text{R}} \tag{5}$$

where, K_{10} is the *intrinsic* toughness of the material without *R*-curve effects, and K_R is *R*-curve toughness, due (in the model cases) to bridging grains in the crack wake. To determine K_{TH} , the *R*-curve toughness is first obtained in the usual manner, followed by a determination of crack dimensions at each point along the *R*-curve by either direct measurement or indirectly imputed using substitutional modifications to the Griffith and Paris equations. (The derivation can be found in the foregoing references.) Using this methodology, Kruzic et al. found K_{10} values of 1.3 MPa m^{1/2} and 1.4 MPa m^{1/2} for Al₂O₃ and Si₃N₄, respectively [388]. Remarkably, their results demonstrate little difference in *intrinsic* toughness between materials that have strongly different *R*-curve behavior. Then, using the K_{10} and K_R values, they derived practical fatigue diagrams for these model materials. The diagram for Si₃N₄ is reproduced in Fig. 11 [388].

Depicted are areas of stress intensity versus crack size where: (1) no SCG is operative under either static or cyclic loading (i.e., below K_{TH}); (2) mixed mode behavior is present, with unstable crack growth under cyclic, but not static loading (i.e., above K_{TH}, but below K_{IC}); and, (3) unstable behavior occurs with either static or cyclic loading (i.e., fast fracture, or above K_{IC}). Later, Greene et al. extended the method to a series of silicon nitrides and reported K_{10} ranging between 2.0 and 2.3 MPa $m^{1/2}$ and K_{TH} falling between 2.6 and 4.1 MPa m^{1/2} [391]. The corresponding K_{TH}/K_{IC} ratios ranged from 0.47 to 0.59. In accordance with predictions, they demonstrated 100% survivability of a limited number of test specimens subjected to cyclic fatigue below K_{TH} for a range of crack sizes from about 60 to 230 µm. The unique feature of this methodology is the combination of R- curves with fatigue data. Their research provides an important summary to this section of the review, with pertinent instructions for practicing biomedical engineers. When designing an implant system which incorporates a bioceramic, predictive service stresses need to be kept below K_{TH} for the reasonable range of inherent flaws present within the ceramic; and, perhaps, more importantly, selection of a bioceramic with high intrinsic (K_{10}) and strong R-curve toughness (K_R) is preferred. A pronounced R-curve

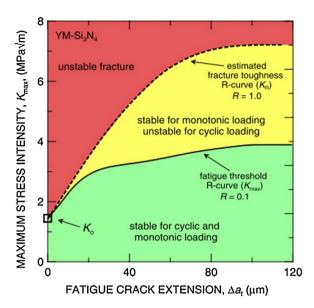


Fig. 11. Fatigue map for YM-Si₃N₄, illustrating regimes of stable behavior for both monotonic and cyclic loading conditions [388]. (YM = Yttria–Magnesia doped). Reprinted with permission.

provides additional safety because short cracks will arrest after a small amount of sub-critical extension rather than progress to failure.

3.7. Biocompatibility

Biocompatibility is an obvious essential requirement for an implantable material. A comprehensive ISO standard has been established for assessing biocompatibility which includes both in vitro and in vivo tests for cytotoxicity, genotoxicity, carcinogenicity, reproductive toxicity, irritation, delayed hypersensitivity, and systemic toxicity. Testing in accordance with this standard is a prerequisite for legally marketed orthopaedic devices in the US and EU [393]. In general, all of the materials shown in Tables 1a and 1b comply with this requirement. However, passing the standard does not guarantee an absence of adverse reactions or potential systemic problems in vivo. A key example is ASTM F799CoCr. While it is employed in multiple implantable applications, its use as both femoral heads and acetabular liners in THA has resulted in considerable morbidity due to tissue incompatibility. CoCr is considered bioinert per the ISO standard principally due to its apparent low solubility. However, the testing protocol does not replicate an evaluation of submicron wear particles which have enhanced dissolution rates. Clinical evidence of the induced harm of these small particles is rapidly growing with numerous adverse events being reported annually. Patients with these devices are advised to have their blood serum levels frequently checked for high concentrations of cobalt, chromium, and nickel which can lead to neurologic, cardiac, endocrine, or ocular impairment [394]. Local formation of pseudo-tumors and tissue necrosis are commonly observed [395,396], and isolated cases of metallosis induced fatalities have been reported [397]. Even for an Al_2O_3 ceramic, which is considered to be completely bioinert, the presence of small wear particles can still produce an inflammatory response. This was aptly demonstrated from in vitro and ex vivo studies by Yagil-Kelmer et al. [398]. They tested the effect of a range of clinically relevant wear particles from 0.5 μm to 1.5 μm on the viability and cytokine expression from human monocyte cells at a particle to cell concentration ratio of 100:1. While no cell lysis was observed, they found that the smallest particles consistently provoked increased generation of cytokines, suggesting that the smaller particles are

bio- incompatible. Another pertinent example is the use of PEEK polymers. Again, these compounds are assessed as being biocompatible, and their performance as arthrodesis devices is acceptable. However, even one manufacturer readily acknowledges that PEEK's osseointegration capabilities are inadequate, resulting in sparse bone on-growth and fibrous tissue formation [235]. These observations have prompted development of Ti- or HAp-coated PEEK or HAp/PEEK composites which effectively employ the added material to solve bone apposition problems [399].

One must remember that the human immunological system is attuned to treating any foreign body as a potential hazard, and prioritizes its defensive actions as: (1) dissolution - if possible, (2) encapsulation – if necessary, and, (3) integration – if appropriate. The physicochemical and physiologic reactions that take place at the implant's surface are complex involving interactions with both inorganic and microbial chemical moieties. Small changes in chemistry at the implant's surface can determine the difference between a device being bio-friendly and one that is rejected. Integration with living tissue is innately difficult for foreign bodies due to a lack of critical biologic functions, such as vascularization, self-reparation, and modification in response to external stimuli [400]. Mechanical loading, friction, and wear all play important roles in determining an implant's biocompatibility and biologic integration. While the goal of any implant is to maintain its function for the life of the patient, all have practical limitations. For articulation devices, present-day failures are due to aseptic loosening, infection, weardebris induced osteolysis, dislocation, or fracture, with the rate of early failures remaining alarmingly high [275]. Remarkably, these same failure modes were recognized by Sir John Charnley for hip arthroplasty in his pioneering procedures [12]. This is perhaps not as surprising as it may seem, given that some popular implant materials of today are essentially the same as those used by arthroplasty pioneers. The body's rejection of an implant typically leads to fibrous tissue encapsulation [401]. The etiology of this type of failure can be due to poor surgical placement or inadequate initial fixation, or it may be associated with the biomaterial's lack of mid- to long-term functional biocompatibility [402]. It is therefore important to have in vitro and in vivo tests that go beyond the general ISO standard. Biocompatibility evidence is basic, but it must also be modified to support the implant's design functions. For instance, biocompatibility of articulation devices needs to include an assessment of the toxicity of fine wear particles; and arthrodesis devices need to be checked for fibrous tissue formation and pseudarthrosis. With this being said, ceramics and ceramic coatings do have biocompatibility advantages over their metal and plastic counterparts. As an example, direct in vivo comparisons of Al₂O₃, ZrO₂, Ti6Al4V, and high density polyethylene (HDP) particles implanted adjacent to murine calvarial bone were performed by Warashina et al. [403]. While all four materials are biocompatible in accordance with the ISO standard, the particles of HDP and Ti6Al4V induced twoto three-times higher osteolytic lesions with significantly greater levels of proinflamatory cytokines than the ceramic materials. Nevertheless, as previously discussed, this does not suggest that bioceramics are wholly bioinert. Indeed, as reviewed by Pezzotti, there are probably no truly bioinert substances [79], with physicochemical changes occurring on all materials in contact with human tissue. However, the biologic reactivity of ceramics is indeed lower, and as described in subsequent sections, this feature can be beneficial.

3.8. Phase composition

In this section, a comparison of phase compositions for ceramics and coatings is provided. This discussion is restricted to analyses of "new" and "retrieved" components. Subsequent sections of the review will discuss effects of hydrothermal and corrosive conditions on phase stability.

Shown in Table 1a are surface phase compositions for ceramics. Two materials, α -Al₂O₃ and β -Si₃N₄ are stable, essentially single phase materials, whereas all transformation-toughened ceramics are at least biphasic and purposely metastable. For Al₂O₃, submicron equiaxed powders are received in their α -phase form. A small amount of MgO is added to the Al₂O₃ powder to inhibit exaggerated grain growth during densification. The α -crystal structure is maintained during sintering, and resulting products consist of $100\% \alpha$ -Al₂O₃; but the presence of MgO, along with raw material impurities generates a residual thin glass phase at grain boundaries [404]. Preparation of dense Si₃N₄ involves an irreversible phase transformation. Similar to Al₂O₃, submicron equiaxed Si₃N₄ powders are received in their α - phase form. With addition of sintering aids, α -Si₃N₄ is 100% converted to its β - polymorph during densification. The α -phase has higher free energy (\sim 30 kJ/mol at 25 °C), which makes the reverse β to α transformation impossible [405]. A thin residual glass or crystallized phase exists at Si₃N₄ grain boundaries. Its presence is dependent on the type and amount of sintering aids and powder impurities [406]. As discussed in subsequent sections, the physiologic environment has minimal or no impact on the phase stability of either α - Al_2O_3 or β - Si_3N_4 . The phase composition of ZrO₂-based ceramics, including Mg-PSZ, Ce- and Y-TZP, ZTA, AMC, and ATZ is composed of tetragonal-, monoclinic- and cubic- ZrO_2 , and α - Al_2O_3 for composites. Processing of these materials begins with co-precipitation or intricate mixing of submicron m-ZrO₂ powders with aliovalent cation dopants (MgO, CeO₂, or Y₂O₃) to form Mg-PSZ, and Ce- or Y-TZP. Separately, admixing submicron α - Al₂O₃ powder in appropriate proportions with TZP powders is performed to obtain ZTA, AMC, or ATZ composites. Other minor additives, such as Cr₂O₃, SrO, TiO₂, and SiO₂ are often included, and these compositions are then densified at temperatures above the m- to t- ZrO₂ phase transition (950-1170 °C). Metastable t-ZrO₂ is predominantly retained upon cooling due to dopant-stabilization and the dense constraining matrix. The data of Table 1a indicate that the surface phase composition of these ceramics varies from 42% to 99% t-ZrO₂, excluding α -alumina for the composites. Mg-PSZ's phase composition after initial firing is approximately 55% c-ZrO₂ with the balance being t-ZrO₂ [70]. The phase composition of as-fired Ce- and Y-TZP is predominately t-ZrO₂ (65-95%), the remainder being m-ZrO₂ and a residual amount of c- ZrO₂ [70].

Excluding α -Al₂O₃, the phase composition of ZTA, AMC, and ATZ ranges from 58% to 99% t-ZrO2, the balance being mostly m-ZrO₂ [162,247]. Minor addition of SrO results in a tertiary magnetoplumbite phase, SrAl₁₂O₁₉, which forms platelet-type grains. Also, as with the other ceramics, impurities segregate at grain-boundaries forming an intergranular glass phase. For AMC, Affatato et al. documented an evolution in the phase chemistry for new, unworn femoral heads over a ten-year period and found chronological reductions in m-ZrO $_2$ from about ${\sim}40\%$ to ~12%, indicating progressive improvements in composition and processing by the manufacturer. Their results are graphically presented in Fig. 12 [247]. Shown also are phase compositions for retrieved, worn femoral heads during the same period. In all cases, increased m- ZrO₂ was observed on worn components. Their results, along with similar findings cited elsewhere within this review, demonstrate the impact physiologic conditions have on the phase evolution of ZrO₂-based bioceramics. The metastability of these ZrO₂-containing materials presents both benefits and risks. Therefore, careful compositional control of the monoclinic content is necessary for enhanced toughness and strength in combination with sufficient hydrothermal resistance, which is seen as a "feature of safety" [160].

Surface phase compositions for PVD ceramic coatings, OxZr, and HAp are provided in Table 1b. The PVD coatings (DLC, TiN, ZrN, and TiNbN) exist as stable amorphous or cubic columnar phases that are epitaxially grown on their respective metallic surfaces

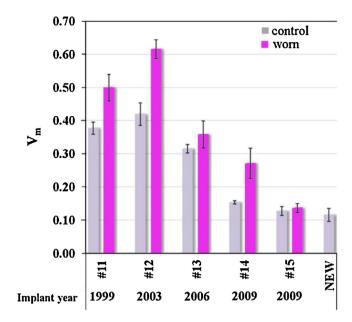


Fig. 12. Average monoclinic volume fraction (V_m) data as obtained from micro-Raman spectra recorded on the unworn (control) and worn areas of analyzed BIOLOX® *delta* femoral heads [247]. Reprinted with permission.

[217,321,407,408]. Even though they are under significant compressive stress because of their growth processes (up to 3.5 GPa), none undergo phase transformation during use [409]. Similarly, OxZr exists predominately as m- ZrO₂ because its oxidation temperature is below the $t \rightarrow m$ transition. However, a small amount of t-ZrO₂ (<5 vol.%) is observed in the microstructure, but does not appear to contribute to any instability [190]. OxZr is also under considerable compressive stress (\sim 670 MPa) due to its growth, and this may contribute to its stability [190]. Finally, as described previously, HAp coatings are multiphase mixtures of CaO, amorphous and crystalline calcium phosphates, and a disordered range of hydroxyapatites that are intentionally designed for dissolution and resorption.

3.9. Corrosion and hydrothermal stability

Corrosive dissolution and hydrothermal degradation of a biomaterial may impact its *in vivo* performance and longevity. On the one hand, debris or dissolved ions can contaminate surrounding tissues leading to formation of pseudotumors or tissue necrosis (*e.g.*, CoCr hard-on-hard bearings), but, on the other hand, controlled dissolution is desirable for other materials (*i.e.*, calcium-phosphates and HAp). Similarly, partial resorption may be preferred for certain hard coatings such as Si₃N₄ [124] but unacceptable for others (*e.g.*, DLC, TiNbN, or OxZr). Consequently, an understanding of each material's corrosive and hydrothermal characteristics is important addenda to basic physical and mechanical properties. In this section, we elucidate on the susceptibility of these bioceramics to chemical corrosion and discuss their hydrothermal stability.

3.9.1. Chemical corrosion

Ceramics are more stable in corrosive environments because they are the oxide or nitride counterparts to metals. While corrosion in metals can be due to electrochemical or galvanic oxidation (*i.e.*, dissolution into soluble ions), ceramics degrade by corrosive attack at grain boundaries. Tribochemical wear can accelerate these processes. Active bioceramics like HAp and TCP degrade *in vivo* by a combination of physicochemical processes and osteoclastic resorption. Most corrosion studies on ceramics have been conducted for industrial applications in severe chemical environments, whereas

Table 3Qualitative corrosion resistance of biomaterials.

Material	Weak acids	Weak bases	Strong acids	Strong bases	Molten salts	Organic solvents	Saline or SBF
Al ₂ O ₃	⊕	Φ	0	0	0	⊕	⊕
ZrO_2	Ø	Ø	\otimes	⊗	\otimes	⊕	+
Si ₃ N ₄	+	+	Ø	Ø	8	⊕	⊕
TiN	+	+	Ø	Ø	\otimes	\oplus	⊕
ZrN	+	+	Ø	Ø	\otimes	\oplus	⊕
440C SS	⊗	⊗	X	X	X	+	+

 \oplus = excellent; += good; \oslash = fair; \otimes = poor; x = very poor; SBF = simulated body fluids.

corrosion in the presence of various body fluids or simulated solutions is less-well characterized. This section reviews the corrosive behavior of these bioceramics in both types of environments. A qualitative assessment of the corrosive behavior of ceramics and selected coatings in acids, bases, organic solvents, molten salts, and saline or simulated body fluids is provided in Table 3. A common medical grade stainless steel (440 °C) is also included for comparative purposes.

For ceramics, α -Al₂O₃ shows excellent corrosion resistance under hydrothermal conditions at neutral pH, in saline solution [410], or in the presence of weak acids or bases [411,412]. It is stable in static saline and body fluids at room temperature [413]. Strong acidic or alkaline solutions [411,412,414,415], high-temperatures, and/or pressures are generally required to initiate significant corrosion [416–418]. Al₂O₃ also exhibits moderate corrosion resistance in the presence of high-temperature molten salts [418]. In all cases, corrosive attack occurs at grain boundaries where impurities such as SiO₂, Na₂O, CaO, and MgO segregate [412], with higher purity compositions showing improved corrosion behavior. Also, minor beneficial corrosion improvements for Al₂O₃ were obtained using nitrogen ion implantation by formation of aluminum nitride at the ceramic's surface [413,419] .

The corrosion characteristics of ZrO₂-based ceramics (*i.e.*, Mg-PSZ, Ce- Y-TZP, ZTA, AMC, and ATZ) are more complex due to variations in chemistry, phase composition, and processing. For instance, Mg-PSZ is severely attacked in dilute acids, whereas Ce-TZP shows improved corrosion resistance, with attack occurring at grain boundaries [420]. Some ZrO₂-based ceramics contain sintering additives such as TiO₂, MgO, SiO₂, or CaO, which partially segregate to grain boundaries [158,421]. Similar to Al₂O₃, these impurities provide a corrosion path from the surface to the bulk [420,422]. Intergranular microcracking associated with the $t \rightarrow m$ transformation permits corrosive solution penetration into the material, eventually leading to disintegration. In fact, hydrothermal phase instability of doped-ZrO₂ is the dominant corrosive mechanism, which is operative at all temperatures and pressures above ambient [73,422].

Si₃N₄ is resistant to attack by common organic solvents and weak acids or bases at room temperature [423,424]. Its corrosion behavior is dependent upon the amount and composition of sintering additives, processing conditions, and the type and concentration of the corrosive media [410,425]. Corrosion is more pronounced in highly acidic or caustic solutions [415,426-432], or in combination with high-temperatures (>300 °C) and pressures (>8 MPa), [433] or in molten salts [434,435]. In one study, sintered Si₃N₄ was more susceptible to attack by HCl than hot-isostatically pressed Si₃N₄, but the opposite effect was found for HF [426]. In two comparable studies, Glauskova et al. found very nearly identical corrosion rates for Al₂O₃ and Si₃N₄ using distilled water at temperatures up to 290°C [410]; and in a third study, Lin et al. found corrosion of Si₃N₄ in NaCl and CaCl₂ solutions to be negligible at room temperature [423]. Similar to other ceramics, corrosive attack occurs preferentially at grain boundaries. In some instances, crystallization of the intergranular glass improves its resistance to corrosive attack [429-433]. In a recent review, compositional tailoring of Si₃N₄'s grain boundaries was advocated by Hermann as the best approach to improving its overall corrosion resistance [436].

Corrosion studies on PVD ceramic coatings typically show favorable results for all organic solvents, most acids, bases, and in hydrothermal conditions [437–444]. Because the coatings possess amorphous or columnar grain structures and are deposited epitaxially from high purity targets, they are devoid of impurities, which favors improved corrosive behavior. Nevertheless, imperfections generated during the deposition process, such as micro-droplets, pinholes, or other point defects become corrosion initiation sites, resulting in wear or delamination of the coating, leading to attack of the underlying metal [445]. As an example, Bolton and Hu studied the corrosive behavior of three biomedical PVD coatings (CrN, TiN, and DLC) on surgical grade CoCr in saline at room-temperature [224]. They found improved corrosion resistance for coated samples, but noted that pit defects continued to expose the substrate to corrosion. They also observed that coating thickness via multiple layers was an effective countermeasure to reduce pit defects. As discussed earlier in this review, the use of multiple interlayers is a method of covering-up defects that may randomly appear in any one layer. A review of corrosion mechanisms and preventive strategies for ceramic coatings has been provided by Antunes et al. [147].

There is a lack of published information on corrosion of in situ grown OxZr. Available data suggest that this ceramic layer is largely free of the defects observed in PVD coatings [180]. The process of forming the layer through oxidation of the base Zr-2.5Nb metal is diffusion controlled, which is self-regulating with respect to thickness. Provided the base metal is free of impurities and surface contamination, and is homogeneous in composition, the oxidized coating will mimic the chemistry and uniformity of the substrate [190]. Corrosion studies have been conducted on a range of Zr–Nb base metal compositions by Zhou et al. [446] and Branzoi et al. [447] with similar findings—increased niobium content improves the corrosion of the metal alloy. It is plausible that the characteristics of the metal may translate to the oxidized layer given that its cation composition is identical, and its microstructure is composed of lath-type m-ZrO2 interwoven with adherent NbO2 or Nb2O5 stringers [190]. It is therefore reasonable to expect that OxZr will have corrosion characteristics similar to other non-transformable ceramics for all organic and aqueous solvents, saline, and bodily

In contrast to the hard ceramic coatings which are corrosion resistant, HAp coatings are intended to degrade. In fact, the most important function of HAp is its controlled dissolution and resorption concurrent with appositional bone formation. As described earlier in this review, thermal sprayed HAp fortuitously contains a multiphase mixture of an amorphous phase, α - and β -tri-calcium phosphates, and several apatite phases, all of which have varying solubility, with the lower order phases being the most soluble. Initial attraction of osteoblasts is generated early in the post-operative period by dissolution of the amorphous phase, whereas long-term boney apposition is governed by HAp. As reviewed by Sun et al. [55], there are six potential degradation mechanisms for HAp coatings: (1) normal dissolution in homeostatic pH; (2) dissolution by

osteoclasts as part of normal bone remodeling; (3) fixation failure resulting in coating delamination; (4) micro-motion abrasion; (5) lamellae cracking due to residual deposition stresses; and (6) preferential attack and dissolution of the amorphous phase. The first two mechanisms comprise the intended function of HAp, whereas the remaining four generate unwanted coating debris. Resorption of the debris can be accomplished by macrophages if they are sufficiently small in size; but beyond about 30 µm, they are reported to be an irritant [55]. Herein lies the key dilemma for HAp coatings: Regulating degradation is essential for adequate fixation, and this is directly related to the process of applying the coating itself. Work performed by a number of researchers suggests that treatments subsequent to deposition or alternative application methods may have advantages in coating chemistry and quality. As examples, Singh et al. demonstrated that post-deposition thermal treatments between 700 °C and 800 °C completely removed undesirable phases [448]. Chen et al. [230] showed a similar result at 600 °C with an associated reduction in deposition defects, and Kwok et al. [449] used electrophoretic deposition and vacuum sintering to improve HAp's dissolution behavior. Further research will undoubtedly lead to the adoption of compositions and methods that balance HAp adhesion with short- and long-term resorption.

3.9.2. Hydrothermal stability

It is critical that the hydrothermal stability of any biomaterial used in joint arthroplasty be investigated and validated. As described earlier, the chemical environment of the human body is such that no material can truly be considered bioinert, even though they may be classified as biocompatible. In this section, studies on the hydrothermal stability of the various ceramics and coatings will be reviewed.

Any discussion of the hydrothermal stability of ceramics should begin with a review of the stability of ZrO2. Its susceptibility to LTD brought to the forefront the need to assess all bioceramics for in vivo deterioration. Unfortunately, introduction of Y-TZP in the mid-1980s occurred without adequate testing and standards. An initial in vitro study suggested that the material had sufficient stability for long- term in vivo use [68], but early failures persisted [74,450]. From 1997 to 1999, two technical reports were published on the material's hydrothermal stability [451,452]. Static testing in water or an autoclave at temperatures from 60 °C to 130 °C were performed to examine transformation kinetics and polyethylene wear in a hip simulator. Hydrothermal degradation was operative under all experimental conditions; but the calculated activation energy for the transformation extrapolated to biologic temperature (37 °C) suggested that the ceramic's deterioration was kinetically limited, and "25-35 years" would elapse prior to m-ZrO2 contents reaching 30% and 40%. At this rate, the researchers concluded that the performance of ceramic-polyethylene wear couples should be unaffected for the lifetime of most patients. Yet, retrievals as well as in vitro studies showed considerable changes in surface finish in some particular heads [77,453]. The apparent contradiction in performance between in vitro and in vivo Y-TZP femoral heads was likely due to higher in vivo stresses which were not addressed nor replicated in the laboratory studies and to variations in the processes from one producer to another, and even from batch to batch. Surface stresses particularly generated during joint laxity (i.e., micro-separation) have been shown to significantly increase wear in THA joints [454,455]. Also, as described previously, a significant number of in vivo fractures from several batches of femoral heads – traced to a processing change – resulted in the September 2001 recall of this material [242]. Although controversial, the LTD mechanism has been described by Sato et al. [456], Guo et al. [457], and later by Chevalier et al. [73]. Addition of trivalent Y3+ ions introduces oxygen vacancies in the zirconia sublattice and induces metastability of the t- ZrO₂ phase. In Kröger-Vink nomenclature, the reaction is as follows:

$$Y_2O_3 \frac{2ZrO_2}{2} 2Y'_{Zr} + 3O_0^x + V_0^{\bullet \bullet}$$
 (6)

where, Y^{3+} ions substitute for Zr^{4+} with a net negative charge of -1(i.e., Y'_{Zr}) and $V_0^{\bullet \bullet}$ represents double positive oxygen vacancies in order to maintain charge neutrality. As long as hydroxyl ions are not present in the environment, metastability of the tetragonal phase is assured due to a reduction in the Zr⁴⁺ coordination number from 8 to approximately 7. However, in the presence of moisture, disassociation of water dipoles into hydroxyl ions (OH) and protonic defects (OH) provides a mechanism for annihilation of vacancies, filling them with either the protonic defects or with oxygen ions and interstitial hydrogen. This destabilizes the tetragonal structure and initiates the polymorphic phase transformation at the surface. As described by Keuper et al. and demonstrated in Fig. 13, the process is autocatalytic, having linear kinetics controlled by the reaction of water with Y-TZP grains. Transformation-induced microcracks at grain-boundaries create diffusion paths which cascade the degradation reaction to subsurface grains, where the process is repeated [458].

Understanding the seriousness of this issue, it is remarkable that little attention was initially given to LTD of ZTA, while the basic mechanism is intrinsically the same. The first reported study was by Insley et al. in 2002 [169]. They examined two ZTAs designated CZTA and NZTA. The composition of CZTA was 74% Al₂O₃, 24% ZrO₂ and 1% of other mixed oxides, while NZTA was a 75%/25% mixture of Al₂O₃ and m-ZrO₂. X-ray diffraction analysis of the dense bodies showed only t-ZrO₂ at the surface of the NZTA, but up to 35% m- ZrO₂ in CZTA. Ageing of both compositions was conducted in an autoclave for up to 5 h at 134 °C. Samples were also placed in Ringer's solution at 37 °C for up to 12 months. Remarkably, they reported no increase in monoclinic content and no observable degradation for either composition. Later, Deville et al. published two studies on the low-temperature ageing of ZTA [205,459]. Their first paper examined a range of Al₂O₃-ZrO₂ compositions from 2.5 wt.% ZrO₂ in Al₂O₃ to pure 3Y-TZP. Ageing experiments were carried out in an autoclave at 140°C under a pressure of 2 bars for up to 115 hours. No ageing was observed for Al₂O₃/ZrO₂ mixtures using unstabilized ZrO₂. However, significant ageing followed by microcracking was noted at Al₂O₃ grain boundaries for ZrO₂ contents of >15 wt.%, resulting in a pathway for water diffusion from the surface towards the bulk. Compositions exceeding the percolation limit exhibited up to 80 vol.% m-ZrO₂ [460]. All Y₂O₃doped Al₂O₃/ZrO₂ mixtures showed hydrothermal degradation, reaching levels of up to 25 vol.% for the highest ZrO2 contents. The most degradation was observed for 22 wt.% 3Y-TZP in Al₂O₃, which corresponded to the percolation limit. Their second study focused on ZTA degradation using only Y- TZP/Al2O3 composites [205]. Compositions from 10 to 20 vol.% Y-TZP were aged for up to 80 hours in an autoclave at between 110 °C to 140 °C. All compositions showed surface transformation even for short exposures. Higher ZrO_2 compositions showed the greatest change, from $\sim 10\%$ m-ZrO $_2$ for sintered samples, increasing to \sim 25% after ageing; and all samples displayed a rapid rise in m-ZrO₂ during initial exposure, followed by a more gradual increase for the duration of the test. No degradation plateau was observed for any composition.

Additional studies by Pezzotti et al. were conducted on commercially available AMC ceramics from 2007 through 2011 [244,246,461–463] . Femoral heads were subjected to autoclave conditions of 121 °C for up to 300 hours. Using Raman spectroscopy, surfaces were examined for changes in ZrO_2 phase composition. They showed that the $t \rightarrow m$ transformation progressed at a high rate during initial exposure (<30 h) and tended to slow thereafter, but did not plateau. At the conclusion of the test, ~75 vol.% of the

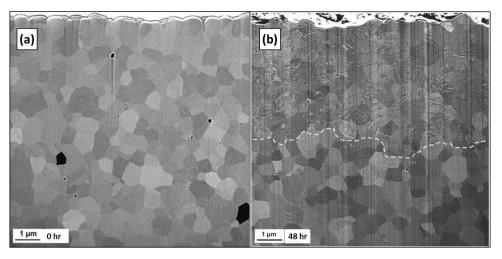


Fig. 13. SEM micrographs of the cross-section of Y-TZP showing progressive evidence of LTD: (a) as-sintered sample with no hydrothermal treatment; and, (b) hydrothermally aged in a saturated steam autoclave at 134 °C and 3 bar for 48 h. Note the clear demarcation between transformed and non-transformed grains and microcracks in (b) [458]. Reprinted with permission.

available ZrO2 at the surface had transformed. They also found \sim 28% reduction in surface fracture toughness associated with the transformation. These studies confirmed that AMC was sensitive to hydrothermal degradation; but this result is not unexpected since transformation-toughening is a key feature of this composite ceramic. It also has to be stated that all these in vitro conditions were severe, and the effects were only surface related, with bulk properties remaining unaffected. There have been very few in vivo failures of AMC femoral heads, and they cannot be correlated so far to LTD. According to one report, approximately 2.2 million devices were implanted by June of 2013, with a reported fracture rate of 0.001% [40]. Retrievals have determined that LTD is operative, but so far, there is no correlation between observed surface transformation and implant functionality [464,465]. Additional research and retrievals will be needed to assess the long-term impact of this phenomenon on the longevity of AMC implants.

It should be noted that Mg-PSZ is also affected by LTD, but its degradation is retarded because of Mg-PSZ's unique microstructure consisting of discrete nano-sized t-ZrO $_2$ precipitates embedded within larger c-ZrO $_2$ grains. Moisture must diffuse from grain-boundaries through the cubic matrix to destabilize the precipitates. Its diffusion coefficient is low, so degradation kinetics are expected to be slow [76]. This was experimentally observed by Swain. Severe autoclave exposure of two Mg-PSZ compositions at 400 $^{\circ}$ C and 6 bar pressure for 4 h showed an increase of \sim 17% m-ZrO $_2$ coupled with a modest decline in fracture strength [466].

Returning to Al₂O₃ or using Si₃N₄ might appear to be safe options in avoiding the observed LTD in ZrO2 based materials. After all, both ceramics are innately phase stable, and presumably immune to attack by moisture. However, as we shall see, this presumption is not entirely correct. Several studies provide insight into the comparative hydrothermal stabilities of Al₂O₃ and Si₃N₄. Sato, et al. examined the hydrothermal stability of a number of Al₂O₃ compositions and Si₃N₄ in caustic alkaline solutions at temperatures up to 200 °C. As expected, they demonstrated no phase change in either bioceramic, and found essentially equivalent dissolution rates for 99.5% Al₂O₃ and hot-isostatically pressed Si₃N₄ [467]. Oda, et al. studied both the hydrothermal corrosion of Al₂O₃ and Si₃N₄ under very severe aqueous conditions (300°C and up to 85 atm) [416,433]. They showed that both materials degraded by an identical mechanism, which involved attack at the grain boundaries, followed by dislodgement of individual grains. The rate of corrosion for Si₃N₄ was higher than for the alumina, but this was not unexpected given the extreme conditions of the test. Conversely, less

severe, and biologically relevant hydrothermal testing was separately conducted on Si₃N₄ by Bal et al. [94], and on Al₂O₃ by Pezzotti [79]. Bal et al. tested dense Si₃N₄ specimens autoclaved at 120 °C in 1 bar steam for 100 h. Components were examined for phase composition and strength both before and after autoclave exposure. There were no observed phase changes, nor were there any differences in the flexural strength (>900 MPa). Using cathodoluminescence spectroscopy on autoclaved samples of Al₂O₃, Pezzotti demonstrated that hydrothermal ageing led to the formation of oxygen vacancies in the alumina lattice. Mechanistically, this is shown in Fig. 14 [79]. Hydrothermal and tribochemical activation of Al₂O₃ results in concurrent hydroxylation of aluminum ions and the release of oxygen along with substitutional inclusion of Mg²⁺ or Ca²⁺ cations forming a spinel-like structure, which is thermodynamically favored over Al_2O_3 [468] . Due to inherently weak grain boundaries and corrosive attack by proton and hydroxyl ions, increased surface roughness and microcracking ensues, which may lead to enhanced wear.

Hydrothermal exposure and/or tribochemical wear of Si_3N_4 also result in marked surface changes. The degradation mechanism is outlined in Fig. 15. As described by Galuskova et al. [410], Dante et al. [469], and also by Pezzotti [79], dissolution of Si_3N_4 occurs by preferential attack of Si_3N_4 bonds, with the release of ammonia (NH₃) and hydroxylated silicic acid ($Si(OH_4)\cdot xH_2O$), forming a tribochemical surface film. The presence of this tribofilm leads to low friction for self-mated Si_3N_4 , particularly under moderate- speed continuous-motion bearings [114,469,470]; whereas frictional spikes (*i.e.*, stick-slip) are observed with slow-speed non-continuous-motion couples [116]. Also, in contrast to Al_2O_3 , which gives up oxygen under hydrothermal conditions, Si_3N_4 is an oxygen scavenger, which may be an advantage in articulation against polyethylene—preventing its premature oxidative deterioration [79].

Few hydrothermal stability studies have been conducted on hard ceramic coatings. At room temperature, these coatings are highly resistant to attack by water, and even offer protective advantages in aqueous abrasive slurries [440]. At higher temperatures and under hydrothermal pressure, DLC films change their carbon coordination from diamond-like to predominately graphitic with a corresponding loss in hardness [471]. In the presence of acids or bases, TiN and ZrN coatings oxidize to form TiO₂ or ZrO₂ at their respective surfaces [409]. However, as discussed previously, the greatest problem associated with these coatings is their discontinuous nature. Localized water penetration and corrosion of

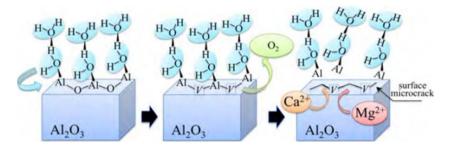


Fig. 14. Tribochemical behavior of Al₂O₃ in a biologic environment [79]. Reprinted with permission.

the underlying metallic substrate is the most common problem *in vitro* [441] or *in vivo* [472]. There are no reported hydrothermal stability studies for OxZr; but this layered ceramic is expected to be highly stable. As discussed previously, perhaps the only concerns for this coating are its thickness and hardness differential between the coating and the substrate, which could lead to delamination [326]. Finally, as presented in the prior section on corrosion, HAp coatings are designed to be hydrothermally active in order to facilitate osseointegration.

In summary, a review of pertinent literature demonstrates that all ceramics and likely all ceramic coatings are susceptible to hydrothermal degradation to a greater or lesser extent. The $t \rightarrow m$ transformation that provides ZrO₂-based materials with enhanced toughness and strength under dry conditions may lead to changes in phase chemistry and properties upon exposure to humid environments. Therefore, under moist conditions, the $t \rightarrow m$ transformation must be assessed during development, or after a process change, and certainly prior to clinical use. Long-term research must also be conducted to assess in vivo transformation rates of new ZrO₂ containing materials. Al₂O₃ and Si₃N₄ are phase stable materials under hydrothermal conditions, but are not totally immune to physicochemical surface changes upon interaction with biologic fluids. Similarly, PVD ceramic coatings and OxZr are also stable, but concerns remain about their protective ability over their metallic substrates due to differential hardness, high compressive surface stresses, and deposition defects. However, it should be noted that corrosion and hydrothermal degradation also occur for metal and plastic implants; and on balance, bulk ceramics and ceramic coatings are innately more stable. Consequently, the most important challenges are development of scientific assessments of each material's degradation mechanisms and kinetics, and

improving the robustness of their respective processes, in order to eliminate potentially negative consequences to patients.

3.10. Hardness and wear resistance

Hardness is a measure of a material's resistance to elastic or plastic deformation under an applied compressive load [473]. It is a fundamental material property that is related to crystal structure, bonding, and bond density, but can be affected by secondary phases, impurities, grain boundaries, and residual stress [474,475]. Ceramics, which are ionic or covalently bonded, have intrinsically higher hardness than metals or polymers (cf., Tables 1a and 1b). Material hardness plays an important role in artificial joints for one primary reason – wear resistance. All metal, ceramic, and ceramic coatings can be manufactured and polished to extremely fine surface finishes for articulation devices (i.e, <20 nm R_a). In vitro hip simulator studies, along with in vivo retrievals, demonstrate that polycrystalline ceramics or hard ceramic coatings are more scratch-resistant [150,258,476-478], typically have lower wear rates, and exhibit less osteolysis for self-mated bearing surfaces [247,479–489] and for articulation against polyethylene [150.476.477.490–499] than do CoCr implants. With less than half the hardness of ceramics, CoCr is particularly susceptible to third-body scratching. A scratch in CoCr produces an abrasive gouge below the nominal articulation surface, and also plastically displaces material above the surface. Both effects lead to accelerated wear [500]. Conversely, because of increased hardness, ceramics are more scratch resistant with shallower depths and smaller surface protrusions.

The impact of hardness on wear resistance can be observed from the comparative wear data presented in Tables 1a and 1b.

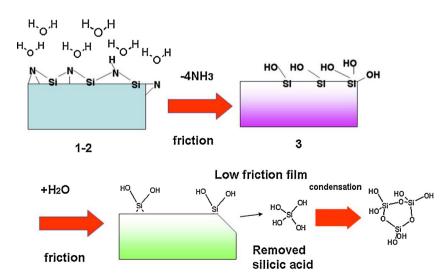


Fig. 15. Tribochemical wear mechanism of Si_3N_4 in a biologic environment in four stages [469]. Reprinted with permission.

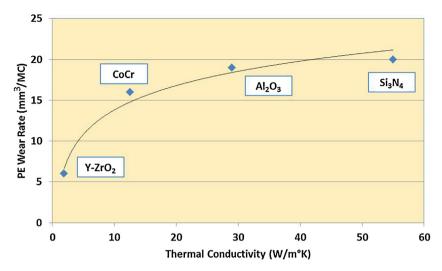


Fig. 16. Polyethylene wear rates versus femoral head thermal conductivity. Adapted from Bowsher et al. [501]. Used with permission.

Wear rates are given for conventional (PE), highly-cross-linked polyethylene (XLPE), and for hard-on-hard bearings using data compiled from hip simulator studies conducted under various protocols during the past 20 years - which partially accounts for the broad range of the observed values. Polyethylene wear is most pronounced for non- or minimally-cross-linked polymers (<40 kGry), and decreases asymptotically for the highly- cross-linked materials (>50 kGry). However, as pointed out by Bowsher et al., any perceived differences in wear need to be tempered with the knowledge that protein precipitation occurs on femoral heads having lower thermal conductivities [501]. This effect is shown in Fig. 16. Less wear is noted for Y-TZP and CoCr in comparison to Al₂O₃ and Si₃N₄ due to greater protein precipitation. Proteins are precipitated because of increased interfacial temperatures. Their presence masks the true wear rate of the bearing. Wear rates for hard-onhard bearings are lower than XLPE, with AMC and ZTA ceramics providing the best wear performance; however, protein precipitation during in vitro wear tests may also play a role in the variability of these wear measurements [502].

Table 1b shows that PVD ceramic coatings or OxZr also provide improved wear resistance because of their high hardness, but only in articulation against polyethylene. Ceramic coatings are unlikely to be effective as hard-on-hard bearings because of the potential for coating failure. Indeed, third-body particles can easily scratch through these thin films, or pit defects can facilitate corrosive penetration, delamination, and catastrophic wear of the underlying metallic substrates. This continues to be a relevant clinical issue as evidenced by several in vivo failures. For instance, a clinical trial comparing DLC-coated Ti6Al4V and Al2O3 femoral heads articulating against conventional polyethylene showed disastrous results for DLC implants due to corrosive pitting and spallation, leading to a revision rate that was four times the Al₂O₃ control group [133,503]. Several case studies have reported similar poor in vivo results for TiN coatings [472,504,505]; and TiNbN coatings were recently found to have completely worn through in another study [506]. OxZr appears to be a solution for these failures due to its substrate coherency and an absence of surface- defects, with clinical results supporting its equivalence to CoCr [144]. However, recent publications have expressed caution over the use of OxZr. Its surface can be easily scratched if: (1) it is malpositioned during insertion; (2) patients experience subluxation or repeated dislocation; or (3) third-body wear particles are present [507–510]. As demonstrated by Lee et al., when an OxZr coating is breached, the underlying soft Zr/2.5Nb alloy can increase polyethylene wear by 161% in comparison to an identically sized CoCr head [150]. These clinical studies and case reports point to the fragile nature of thin hard coatings, and suggest that further development is necessary to improve their reliability. Under ideal laboratory conditions their performance is remarkable; but given on-going *in vivo* failures, clear and convincing clinical evidence will be needed to overcome growing prejudice against their use. Conversely, bulk ceramics provide all of the benefits of thin films with none of the consequences; and the current risk of *in vivo* fracture has become inconsequential. In summary, bulk ceramics should be considered as articulation couples over hard ceramic coatings due to their scratch resistance and low wear, either as self-mated bearings or in articulation against polyethylene.

3.11. Bacteriostasis

Peri-operative or latent infections are a leading cause for hip, knee, and spinal revision surgery [273], with a rate of 2.7–18% [511]. Their occurrence is devastating for patient and practitioner alike, typically requiring surgical removal of the implant, debridement of surrounding tissue, extended hospitalization for antibiotic therapy, followed by re-implantation of an entirely new device [512]. The economic impact to treat infections is typically $4\times$ the cost of primary arthroplasty [513] . While the burden to the healthcare system is large, degradation of the patient's quality of life during treatment, revision, and convalescence has to be considered equally distressing. Nowadays, because of this problem, developing implants with natural or engineered anti-infective properties is garnering considerable attention. Recent reviews have discussed strategies for developing antibacterial devices [514,515]. Interest has focused on preparing composites which perform a dual function, serving both as a structural member (e.g., spinal fusion cage, acetabular cup, coated femoral stem), and incorporating an elutable bactericide (e.g., silver, zinc, copper, iodine, vancomycin, etc.) to prevent biofilm formation or kill planktonic bacteria [516,517]. Elutable calcium from an antimicrobial soda-lime glass-ceramic containing combeite and nepheline crystals was shown by Cabal et al. to effectively resist biofilm formation from five nosocomial strains, while demonstrating excellent biocompatibility towards mesenchymal stem cells [518]. An alternative strategy is to functionalize the implant's surfaces with non-leachable compounds that are contact biocides. Such compounds can be natural or synthetic polymers and include chitosan, peptides, quaternary amines, and N-halamines [514,515]. These and similar "active" strategies contrast with more "passive" methods which rely on the material's inherent antibacterial features. Bacteriostasis is the term used to describe this behavior, and is defined as the property of a material which limits attachment and growth of bacteria without necessarily killing the microorganism. Differences between bactericidal and bacteriostatic materials are significant. Bactericidal compounds often result in large bacteria kill ratios of >99%, whereas bacteriostatic materials only prevent biofilm formation, with reductions from 10% to 90%. There are standardized tests for assessing the effectiveness of implants incorporating bactericidal compounds [519–521]; but, with the exception of a test for plastics [522], none currently exist for comparing bacteriostasis.

In general, ceramics and coatings are considered bacteriostatic. Unlike polymers and metals, they inherently resist biofilm formation. A number of bacteriostasis studies have been conducted on ceramics for arthroplasty and dental applications. Ginebra et al. [523] examined Al₂O₃, Si₃N₄, and AMC in vitro for biofilm formation using two bacterial strains (Staphylococcus epidermidis and Escherichia coli). In comparison to a polystyrene control, all three ceramics showed between 30% and 60% less biofilm formation. In a separate study, Hizal et al. prepared nanostructured and hydrophobic Al₂O₃ and showed that it was effective in repelling Staphylococcus aureus and Escherichia coli [524]. For dental applications, Yamane et al. completed an in vivo human oral study examining titanium, ZrO₂, Al₂O₃, and HAp, all with uniform surface roughness, against Au-Pt alloy as the control. They found no differences among the ceramics as compared to lower adhesion on the Au-Pt alloy [525]. In contrast, Yoshinari et al. demonstrated that Al₂O₃ coated titanium oral implants were significantly more effective in resisting biofilm formation than titanium alone [526]; and Al-Radha et al. found similar results for ZrO2-coated titanium or ZrO₂-abraded titanium in oral studies [527]. A contrasting bacteriostasis study on several grades of Y-TZP was performed by Karygianni et al. in vitro using Enterococcus faecalis, Staphylococcus aureus, and Candida albicans. They observed no anti-adhesive improvement of the ceramics when compared with bovine enamel slabs as controls [528]. Conversely, Gorth et al. tested Si₃N₄, titanium, and PEEK with five separate nosocomial bacterial strains in vitro and consistently found lower amounts of biofilm formation (up to 90%), and fewer live bacteria on Si₃N₄ than either of the other two biomaterials. As might be expected, PEEK generated the highest amount of biofilm and harbored the most live bacteria. They postulated that reasons for the observed differences were likely due to: (1) improved hydrophilicity of the ceramic in comparison to the moderately and highly hydrophobic metal and polymer, respectively; (2) chemical moieties on the surface of Si₃N₄ including charged amino and silanol species of SiNH₃⁺, SiOH₂⁺, SiO⁻, SiNH₂, and SiOH, all of which may act to repel microbial attachment; and (3) a micro-textured surface for Si₃N₄, which may also prevent biofilm formation. These features were not observed with the other two biomaterials [107]. Webster et al. pursued this work using an in vivo Wistar rat model. Samples of Si₃N₄, PEEK, and titanium were pre-inoculated with 10⁴ S. epidermidis prior to implantation into the animal's calvaria followed by incubation for up to 3 months. Provided in Fig. 17(a) through (d) are histological analyses conducted subsequent to animal euthanasia, showing a latent infection on the surfaces of the PEEK and titanium implants, whereas none was present for Si_3N_4 [106].

A similar range of results has been observed with the hard ceramic coatings. For instance, a dental study by Grobner-Schreiber et al. using ZrN indicated significant reductions in oral bacteria compared with uncoated samples; but the diversity of the bacterial flora were not affected [529]. For DLC, Love et al. reviewed a number of bacteriostasis studies which showed that hydrophobic carbon coatings with low hydrogen contents were very effective as a bactericide, even when compared with the antibiotic *Gentamicin* [148]. Finally, Shida et al. performed an *in vitro* bacterial adhesion study for OxZr, CoCr alloy, titanium, and stainless steel implants

using *S. epidermidis*. All samples were carefully prepared to have similar surface roughness. Reported amounts of *S. epidermidis* that adhered to OxZr and CoCr were significantly lower than for titanium or stainless steel (up to \sim 60%) [530].

Clearly, the range of observed results for both the ceramics and coatings suggests that exogenous factors play key roles in determining bacterial adhesion. A review of potential parameters governing biofilm formation was conducted by Renner et al. [531], which included physical, chemical, and interactive factors between orthopedic devices and bacteria.

They concluded that interactions between microbes and biomaterials are complex, and the diversity and adaptability of differing bacterial strains to the milieu of the human body make it difficult to construct even simple engineering guidance. For example, it is generally accepted that hydrophilic substrates are more resistant to biofilm formation; but this is not true for all microbes, with common S. epidermidis preferring either polar or hydrophobic surfaces [532]. Certainly, the unique physical and chemical environment presented by an implant's surface is important. However, while surface roughness, topography, and chemical properties may dictate bacteria attraction, it is the external functions of the microbe itself (i.e., flagella, fimbriae, and pili) which facilitate its attachment. In this regard, bacteria have an advantage because of their living nature. They have motility, selectivity and are adept at exploiting weaknesses present in inanimate implants or the surrounding environment. Not surprisingly, defensive strategies against their onslaught may need to be as varied as the bacteria themselves. Potential solutions include dense implants coated with bactericidal or bacteriostatic compounds, porous structures infused with elutable antibiotics, or devices engineered with micro- or nanostructured features that resist biofilm formation. These and other strategies are active areas of research. In summary, development of anti-bacterial bioceramics represents a new frontier—one in which necessary structural functions are combined with bacteriostatic or bactericidal features to combat an evolving variety of nosocomial microbial strains.

3.12. Osseointegration

The rigid and permanent fixation of an alloplastic medical device with native bone is referred to as osseointegration (or sometimes as osteointegration). Obviously not all bioceramics and coatings are designed for integration with bone. Some are solely important for articulation and wear. However, at the end of the day, implantable medical devices will ultimately fail if they are not soundly supported by the musculoskeletal system. In this regard, osseointegration is as critical in its function as wear resistance is in its respective role. While significant understanding has been acquired with regards to wear resistance, our knowledge of factors affecting osseointegration is only now coming to light. Interactions between the biological environment and implant surfaces are complex, involving a sequence of events that begins with protein absorption and, if successful, ends with fixation via bone on-growth or in- growth. Previous generations of ceramics were considered efficacious if they were merely classified as being bio-inert or biotolerant, meaning that they did not elicit an inflammatory response. However, this does not suggest there were no reactions. On the contrary, the body's immunological answer to bioinert materials has typically been isolation and fibrous tissue encapsulation. However, to effectively integrate, bioceramics must go beyond this simple one dimensional classification. They must be bioactive and/or bio-resorbable, capable of interacting with tissue, inducing bone formation, and providing fixation.

The biological sequence of events leading to osseointegration is complex and dependent on factors associated with the physicochemical structure of the implant and its interaction with proteins

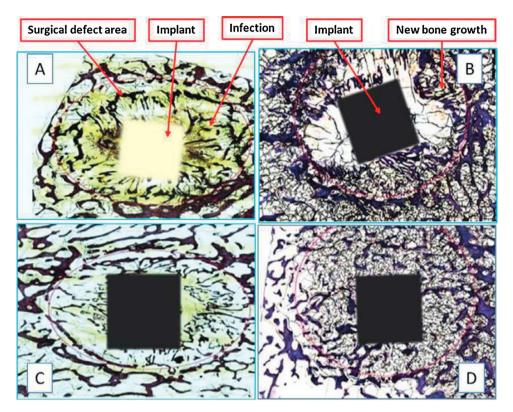


Fig. 17. Effect of biomaterial on latent infection in Wistar rat calvarial defects [106]. Implants were pre-inoculated with 1×104 Staphylococcus epidermidis. Histology performed 3 months after implantation: (a) PEEK; (b) Si₃N₄; (c) titanium; and (d) Si₃N₄. Reprinted with permission.

and living cells. The first and immediate biologic reaction to a foreign body is its coating with proteins. For bioceramics, this process has been documented by Wang et al. in detailed review [533]. Fibronectin, a protein found in plasma binds almost instantaneously to all biomaterials, but other proteins including vitronectin, laminin, collagen, and fibrin also quickly adsorb and participate in wound healing [534]. Adsorption of these molecules is induced by the roughness, hydrophilicity, charge, and topography of the biomaterial's surface, with micro- or nano-structured surfaces being the most effective. Proteins orient themselves to local variations in surface charge and topographical features, and adhesion occurs by electrostatic and mechanical means. Specific active sites on these molecules are recognized by mesenchymal stem cells, mainly integrins, which subsequently bind to the proteins. However, as discussed by Kieswetter et al., cell attachment is neither predestined nor immediate. It may take hours to occur [535] . Again, surface chemistry and topography of the implant dictate this timeline, and remarkably, cells can distinguish between different chemistries and surface features [536,537]. As reviewed by Anselme et al., many studies suggest that surface topography may be more important than chemistry, with more mature and differentiated cells adhering to micro- and nano- structured surfaces [538]. However, the presence of soluble calcium phosphates, bioavailable silicon, and magnesium also appear to be key chemical species promoting adhesion and proliferation of progenitor cells, facilitating their eventual differentiation into osteoblasts [539–541]. Provided the chemical environment remains conducive to continued cell function, osteoblasts secrete a calcified osteoid which leads to endochondral ossification and eventual fixation of the implant. Most of our knowledge with respect to the interaction of cells and biomaterials comes from studying titanium alloys, but features that promote its osseointegration appear similar if not identical to those that are likely operative for bioceramics [542-544]. Ceramic scaffolds possessing interconnected porosity have also been shown to

be effective in boney in-growth for a number of different materials including bio-glasses and ceramics [545], tantalum [546,547], PEEK [548], and of course, various HAp compositions [60,549]. The integration ability of these scaffolds is dependent upon surface chemistry and the size distribution of engineered porosity. As with on-growth, a chemical environment with appropriate concentrations of Ca, P, Si, and Mg are essential for in-growth, along with porosity that is within a preferred range of $\sim\!100~\mu m$ to $\sim\!500~\mu m$ [550]. Although integration occurs for pores smaller than $100~\mu m$, vascularization of the porous structure is most pronounced when openings between the pores exceed this minimum [550] .

A number of the ceramics and coatings have been tested for their ability to interdigitate with living bone, beginning with porous Al₂O₃ in the 1970s [21]. While this early work showed that ossification is possible, the immunological response to Al₂O₃ at that time was encapsulation in fibrous tissue. Bioactive coating strategies have subsequently been employed for Al₂O₃ to alter its local chemical environment and thereby improve boney apposition [551,552]. The use of Y-TZP for dental implants has spawned numerous studies examining its osseointegration characteristics with a review performed by Hisbergues et al. in 2009 [80]. While significant human clinical data are lacking, in vivo animal studies suggest that its osseointegration performance is at least equivalent to titanium alloys (i.e., the gold standard for oral surgery). Effective osseointegration has also been demonstrated for mixtures of ZrO2 and Al₂O₃. For example, using push-out strengths as a measure of integration, Burgkart et al. examined AMC in an ovine model [553] . They showed that a pre-engineered macro-porous structure required approximately eight times the push-out force when compared with smooth implants, and histomorphological evaluations revealed consistent and substantial boney apposition within the porous structure at 12 weeks post-operatively. Similar effective osseointegration for ATZ dental implants has also been reported [554] . For Si₃N₄, several studies demonstrated its osseointegration capabilities. Anderson et al. showed osseous penetration of up to 3 mm in $\sim\!\!72\%$ porous Si_3N_4 with excellent vascularization twelve weeks after implantation in an ovine model [105]. Similar studies by Guedes e Silva et al. showed good boney apposition for Si_3N_4 samples in rabbits evaluated eight weeks after implantation [103,104]. In Wistar rats, Webster et al. compared the osseointegration of Si_3N_4 , PEEK, and titanium, reporting that Si_3N_4 had improved osseointegration over the other two biomaterials at all-time points up to 3 months post-operatively, both under aseptic conditions and when implants were pre-inoculated with 10^4 S. epidermidis [106].

There are few reported studies on the osseointegration of hard PVD ceramic coatings. As discussed previously, while these coatings are biocompatible, their ability to fixate to bone is governed by local surface chemistry and topography. From a chemical standpoint, they are largely bioinert, and their surface roughness is determined by the underlying substrate. Consequently, improvements in osseointegration are expected via topography modifications of the substrate or by adjustments to deposition composition. Recent studies suggest that some or all of these strategies may be effective. For example, DLC coated titanium implants showed significantly greater boney apposition and bone thickness growth than comparable bare titanium in an in vivo femoral study conducted on rats [555]. An in vitro study using DLC suggested that several compositions may provide osseointegration improvements [556]. Alternatively, for TiN coatings, Sovak et al. found excellent mineralization at the interface of implants in Wistar rats, and concluded that TiN was at least as effective in boney fixation as uncoated Ti6Al4V substrates [557]. Finally, for dental implants, Rizzi et al. found that application of a ZrN coating to titanium markedly improved osseointegration and reduced healing time [558]. Similar results might be expected from the other ceramic coatings, but data are currently lacking.

As might be expected, studies on the osseointegration of HAp coatings have been numerous and significant - and no more so than for those that have examined the long-term survivability of HAp-coated hip stems [49,52,559,560]. A number of these studies report Kaplan-Meier survivability of >92% at 20+ years post implantation. Failures were mostly unrelated to the hip stem itself, and either involved anatomical changes, wear, osteolysis, and aseptic loosening of the acetabular cup, or other medical problems. Patient satisfaction with these implants remained high (also at 92%) at final follow-up, with essentially all femoral stems showing signs of stable fixation. Development of HAp coatings is closely related to parallel innovative research into porous synthetic bone scaffolds. Osseointegration is an essential requirement for these devices as well. Today, autograft and allograft are primarily used for rectifying large bone defects due to trauma, tumor resection, osteomyelitis, or necrosis. However, limitations which preclude their wide- spread use include restricted availability, donor site comorbidities, and risks for infection or disease transmission. Consequently, ceramics, glasses, and orthophosphate composites are being extensively investigated as alternative bone-building therapies. The interested reader is referred to a number of recent relevant reviews in this field [44,60,561-567]. There is general agreement that an ideal bone scaffold must be biocompatible and osteoinductive, having a broad range in porosity for cellular proliferation, neovascularization, and bone ingrowth, possessing a strength greater than the bone being repaired, but having a similar elastic modulus to prevent stress shielding. Furthermore, the scaffold must be customized in properties and shape to closely match the osseous defect, and be slowly resorbable concurrent with the ingrowth of new bone. These are challenging requirements given differences in patient age, bone quality, and overall health. HAp and tri-calcium phosphate are commonly used clinically because of their demonstrated biocompatibility, bioactivity, osteoinductivity, and chemical similarity to the apatite phase of living bone. Nanocrystalline HAp particles, fibers and composites having open interconnected porosity, from about 100-900 µm, compounded with bioglass (i.e., 45S5 or β-wollostonite) or synthetic polymers (e.g., polymethylmethacrylate, polyacrylic acid, polyvinyl alcohol, and others) display increased mechanical properties, good biocompatibility, and bioactivity [58,60,566]. Other organic-inorganic multicomponent HAp scaffolds are being explored, including biomimetic adsorption, or seeding of the porous structure with collagen, gelatin, chitosan, growth hormones and factors, bone morphogenetic proteins, genes and mesenchymal stem cells, or various therapeutic drugs. These additives are designed to induce osteo- and angiogenesis, or target disease and prevent infection, respectively [568]. While many of these strategies are exploratory in nature, they demonstrate the breadth of research being conducted on HAp composites, with more than 10,000 scientific publications occurring within the last 4 years [569]. It spite of this progress, utilization of synthetic bone scaffolds remains in its infancy. Challenges that must be overcome include: (1) continued improvement of mechanical properties, particularly fracture toughness, through the use of nanostructured materials, and by developing ceramic-ceramic, ceramic-glass, or ceramic-polymer composites; and (2) enhancement of bioactivity through incorporation of the aforementioned growth factors, genes, cells, or biological agents. Ultimately it should be possible to engineer a new generation of bespoke-based HAp scaffolds to address specific patient pathologies [44]. With attendant developments in bone tissue engineering, the effectiveness of HAp coatings in total joint arthroplasty will undoubtedly improve as well.

To summarize, osseointegration mechanisms for ceramics and ceramic coatings are complex, but dependent principally on two interrelated factors: (1) the biologic environment next to the implant itself, including its hydrophilicity, surface charge, and availability of osteogenic elements (e.g., Ca, P, Si, and Mg) and compounds (e.g., collagen, chitosan, synthetic biocompatible polymers, and therapeutic drugs); and (2) the structure, topography and roughness of the implant's surface, or the size distribution of its pores and connecting channels. These engineered features promote protein adhesion and attract progenitor cells which differentiate into osteoblasts and osteocytes, leading to the firm fixation of orthopaedic devices.

4. New directions in bioceramics

4.1. Novel ceramic materials and composites

Over ten years ago, Dr. Larry Hench, the distinguished researcher and inventor of 45S5 bioglass [570], forecasted the development of a third generation of biomaterials [571]. Whereas the first two generations involved use of bioinert and then bioactive or resorbable materials, respectively, he envisioned a third generation designed to assist in healing the body. His prognostication is becoming a reality, but not by abandonment of first and second generation materials. Indeed, technological progress on all three generations exists on a continuum and are additive. In the following sections, we sequentially discuss new directions in bioinert, bioactive, or resorbable ceramics, and reiterate Dr. Hench's view with respect to the future, which includes "smart ceramics."

4.1.1. Bioinert ceramics

Fifty years of clinical experience has demonstrated that there are no truly bioinert materials. All induce an immunological response, but ceramics tend to minimize this interaction. Bioinert ceramics have evolved through trial and error experimentation from Al_2O_3

(modest strength) to ZrO₂ (high strength, but also high sensitivity to LTD) to ZrO₂-Al₂O₃ composites (high strength with improved LTD resistance). While the current standard in bioinert ceramics is BIOLOX® delta (CeramTec, Plochingen, Germany), its position may eventually be challenged by other ZrO₂ and ZTA compositions having both high strength and little or no LTD, or by non-oxide ceramics such as Si₃N₄ with elevated strength and no LTD. As an example of an improved ZrO2, Matsui et al. demonstrated that addition of a small amount of germanium oxide (0.30 mol%) to a 3 mol% Y₂O₃-doped ZrO₂ coupled with low-temperature sintering produced a fully dense nano- crystalline Y-TZP (grains <300 nm) which exhibited no LTD for 4 years in a steam autoclave at 140 °C [572]. As other examples, ZTA nano- composites which exclude the use of yttria, or which incorporate zirconia nano-particles (<500 nm) evenly dispersed within alumina grains have been developed [154,573]. Both composites result in high strength, tough, and highly LTD resistant ceramics. Torrecillas et al. predict that further developments in nano-structured ceramics and composites should result in a new generation of biomedical implants with lifespans matching the life expectancies of their recipients [574]. Combining Y-TZP with Ta, Ti, or Nb metals to form cermets has also been investigated with promising results. These composites demonstrate improved ageing behavior and mechanical stability, reduced friction against polyethylene, and excellent biocompatibility [575–577]. Even returning to MgO-doped PSZ is a consideration, given its reasonable strength, enhanced toughness, and LTD resistance [578]. Lastly, use of non-oxide Si₃N₄ is also a viable option because it has the requisite strength and toughness, and is absent of any LTD [94].

Overcoming the deficiencies of hard ceramic coatings is also being explored. Recall that thickness, inherent deposition defects (i.e., pinholes and delamination), and wear debris limit the articulation life of these materials. As pointed out by Alakowski et al., the survivability of any thin film under load is proportional to the square of its thickness [579]; with thicker coatings (particularly those composed of multiple interlayers) reducing the likelihood of defects [147,317,580]. Coating thicknesses will need to increase from less than about 5 µm (current state) to greater than 15-20 µm (future state) in order to achieve the goal of being lifetime prostheses. Recognizing a renewed potential for hard surface coatings, a European consortium was formed in 2013 to develop thick PVD Si₃N₄ coatings on cobalt-chromium alloys with a stated objective of introducing them as hip or knee prostheses in clinical trials by about 2018 [581]. This particular bearing surface has the unique advantage of eliminating osteolytic joint debris because Si₃N₄ wears via a tribochemical reaction, the products of which are soluble and resorbable [123,124]. With these ongoing developments, bioinert ceramics and coatings are expected to remain as favored materials for future joint arthroplasty procedures.

4.1.2. Bioactive and resorbable ceramics

Engineered transitions between bioinert and bioactive ceramics represent a new field of endeavor, with advances in the areas of functional gradient materials (FGM) and self-assembled monolayers (SAM). Additionally, while calcium orthophosphates are the standard for bioactive and resorbable ceramics, structural and compositional improvements are enhancing their performance as well.

Functional gradient materials—FGMs are designed by changing the structure or composition within a component to provide variable properties. As discussed by Bahraminasab et al., joint arthroplasty research often tends to be compartmentalized, with some investigators focused solely on bioactivity for improved implant fixation, ignoring problems such as adverse wear or stress-shielding; whereas others concentrate on mechanical properties or friction to reduce stress shielding or wear-induced osteolysis, respetively [582]. Perhaps an optimum solution is introduction of

functionally graded materials which address all three issues. For instance, one side of a ceramic acetabular monoblock cup can be engineered to have high porosity, low elastic modulus, and a bioactive coating (e.g., HAp) to promote osseointegration and minimize stress shielding; while the opposite side has low porosity, high elastic modulus, and hardness for wear resistance. Furthermore, such a device has the added advantage of conserving native bone due to elimination of the metallic shell. This idea has proceeded beyond the conceptual stage. Dual porosity ceramic implants are particularly advantageous in spinal fusion [583,584] and are actively being developed as THA acetabular components [585,586]. FGMs based on composition are also being developed. For example, Laurenti et al. designed an innovative reconstructive implant that combines bioactive 45S5 glass fibers and polyurethane. The distal end of the implant is predominately polyurethane, and serves as replacement cartilage, while the proximal end, which has a high concentration of glass fibers, osseointegrates for fixation. They demonstrated excellent performance in an animal model [587]. In another less demanding example, Zhao et al. functionally distributed iron oxide within a dental ZrO_2 to generate a gradient in color [588].

Self-assembled monolayers—use of self-assembled monolayers represents another effective method of bridging the gap between bioinert ceramics and biologic tissue. In fact, SAMs can be considered functional gradient materials themselves, but on the nanometer scale. The technique involves grafting or binding organo- metal molecules to an implant in a single layer in order to activate an otherwise bioinert surface. A recent review of SAM methods as applied to ceramics was prepared by Boke et al. [589]. They focused on SAMs consisting of organo-silane polymers whose silane heads proximally attach themselves covalently to the ceramic's surface, allowing for further functionalization of their distal tails. Silane-based SAMs are effective because of their ability to readily bind with silicate surfaces. However, other inorganic moieties are also available, including thiolates [590] and phosphonates [591]. The phosphonates are particularly valuable in functionalizing many materials including bioinert ceramics and bioactive calcium orthophosphates, with at least one start-up company commercializing their use [592]. The distal tails of the SAM can be functionalized with other molecules to customize surface properties for hydrophilicity or hydrophobicity, for cell adhesion, to selectively promote or repel protein adsorption, and to resist or kill adherent bacteria [591,593,594]. Another SAM, applied to polyethylene, has been shown to favorably reduce friction between acetabular liners and metal femoral heads in total hip arthroplasty. This unique SAM, consisting of 2-methacryloyloxyethyl phosphorylcholine (PMPC) provides hydrophilicity and effective hydrodynamic boundary lubrication between the two articulation surfaces. PMPC grafted liners dramatically reduced polyethylene wear debris by 99.9%, making their performance equivalent or even superior to ceramic-on-ceramic bearings [595]. Three-year clinical and radiographic outcomes for a 76 patient trial using PMPC grafted XLPE liners and CoCr femoral heads showed very high Harris hip scores (95.6), no adverse events, and mean femoral head penetration rates of <0.002 mm/year at the latest follow- up. This is a remarkable achievement given that comparative trials without PMPC grafted liners showed at least an order of magnitude higher penetration rates—from 0.01 to 0.06 mm/year [596]. While the study is still in its early stages, its positive outcomes portend a bright future for this and similar SAM technologies.

Calcium orthophosphates—hydroxyapatite and various calcium phosphates are already considered bioactive because of their osteoinductive and resorbable properties. Yet, these features are being enhanced by elemental substitutions, development of composites, biomimetic processing, and introduction of nanostructures, with several recent exhaustive reviews highlighting these efforts [44,566,597,598]. While pure orthophosphate mixtures are

effective bone inducing agents, they lack other trace elements commonly found in native bone. Consequently, investigators have doped HAp with silicon [539], fluorine [599], strontium [600], magnesium [601], and carbonates [602] among others in attempts to increase bioactivity and osteoinduction. Additions of bone morphogenetic proteins and growth factors [568], mesenchymal stem cells [603], and therapeutic genes [604] have also been explored for similar reasons. Elutable silver [605,606] or antibiotics [607,608] have been incorporated to inhibit nosocomial infections. Composites of synthetic bio-erodible polymers [609] (e.g., polylactic acid, polyglycolic acid, and polylactic-co-glycolide), natural occurring polymers [610-612] (e.g., alginate, collagen, chitosan), or nonresorbing polyurethane [613] have been constructed principally to improve mechanical properties. Similarly, HAp has been admixed with bone cement and antibiotics for fixation and infection prevention, respectively [614]. Magnetic compounds [615], electrically conductive compositions [616], or carbon nanotubes [617] have been added to HAp with favorable osseous outcomes. Additionally, biomimetic nanocrystalline HAp coatings are among the latest developments that show significant potential. Nano-structures have enhanced mechanical properties, consistent resorption kinetics, and increased osseointegration in comparison to thermally applied coatings [58,261,618]. These various mimeses or modifications have been effective, although few are in clinical use at this time. Each of these concepts will continue to be active areas of research, with future commercial products undoubtedly improving outcomes for joint arthroplasty patients.

4.1.3. Smart ceramics

Smart materials are defined as having adaptable properties that respond to external stimuli (i.e., stress or strain, local chemistry, pH, moisture, temperature, and electric or magnetic fields) [619]. Non-ceramic examples include shape memory alloys and polymers which are designed to expand or contract in response to temperature fluctuations, or to swell or shrink due to changes in moisture or pH. For ceramics, piezoelectric compounds (e.g. barium titanate, BaTiO₃) which generate a voltage under an applied stress are classic examples. Titanium dioxide (TiO₂) is another smart ceramic. Its photocatalytic behavior is effective for self-cleaning glass, water purification, and in prevention or eradication of biofilms [620,621]. Even ZrO₂-based ceramics (i.e., Y-TZP, ZTA, AMC, and ATZ) and Si₃N₄ are arguably smart biomaterials. They dynamically respond to localized stress and strain changes to minimize catastrophic fracture risk. In ZrO₂, the tensile stress field in front of an advancing crack spontaneously triggers the $t \rightarrow m$ transformation, thereby reducing stress and arresting crack propagation. A different microstructural mechanism is operative in Si₃N₄ (i.e., crack wake $bridging)\,which\,yields\,a\,similar\,result.\,These\,smart\,mechanisms\,are$ effective in ensuring the reliability and longevity of bioinert prostheses. However, the third generation of smart ceramics (according to Dr. Hench) needs to further interact with the human milieu to facilitate healing, not just replace defective or diseased tissues. In keeping with this challenge, several recent reviews provide perceptive viewpoints on this new generation of smart materials. Parvizi et al. envision the use of "self-protective" implants, possessing surfaces with permanent covalently-grafted antibiotics or bacteriostatic molecules [622]. They argue that such moieties will be essential in combating a rising incidence of periprosthetic joint infections (PJI) which result in implant loosening, revision surgery, and replacement. Other researchers have taken on this challenge and effective surface coating strategies against PJI are being realized [515,623,624]. Barrere et al. suggest that biomaterial assisted tissue engineering is in its infancy, and hierarchal constructs are the key to its success [625]. Devices must meet at least three criteria: (1) be recognized and accepted by human tissue; (2) restore tissue utility; and (3) stimulate biological function. Separate pertinent reviews by Wegst et al. [626], Perez et al. [627], and Porter et al. [628] discuss the future of biomaterials by drawing from bioinspired, naturally occurring compounds and structures. They predict synthetic composites will eventually be able to imitate living tissue. To do so, the physical, mechanical, chemical, and biological properties of these composites will need to be engineered (internally and externally) to address unique patient pathologies. Adjustments to mechanical properties (i.e., elastic modulus and strength), tailoring of the morphology and biomimetic functionalization of surfaces with grafted polymers, and inclusion of therapeutic drugs, proteins, or stem cells are all suggested strategies. Finally, Holzapfel et al. promote a multidisciplinary approach in which biologists and biomaterial scientists collaborate to advance beyond biocompatibility and bioactivity, to biomimicry—eventually imitating the extracellular matrix (ECM) of natural tissues [629]. Their stepwise vision is shown diagrammatically in Fig. 18.

While they recognize a long road lies ahead in understanding ECM's complexity, they suggest that this knowledge will ultimately achieve the objective of translating smart biomaterials from the "bench to the bedside." Clearly, one ceramic, HAp, is at the forefront of meeting many of the "smart" criteria outlined by these prognosticators. It can be compositionally modified to resist infection. It can include progenitor cells and genes which provide signaling instruction to in vivo cells to differentiate and induce ossification. It can incorporate therapeutic drugs, natural or synthetic polymers, and can be biomimetically deposited to induce native bone growth. Finally, its dissolution kinetics can be tuned so that metabolic resorption occurs consistent with bone reformation. In 2010, a European consortium was formed to explore and conduct clinical trials for a number of these therapies using orthophosphate scaffolds [630]. It is apparent that the modulated features of HAp are living up to Dr. Hench's prediction of a future in which biomaterials aid in repairing instead of replacing living tissue. Although significant research remains, the paradigm shift now occurring with HAp should ultimately be transferrable to other bioceramics.

4.2. Manufacturing processes

There are three general methods for manufacturing ceramic components-subtractive, near-net-shape, and additive. Subtractive manufacturing is the dominant and traditional method. It involves forming of oversized parts followed by material removal to achieve final shapes. Forming methods include dry- or isostaticpressing, slip-casting, gel-casting, extrusion, and injection molding. Machining is performed on pre-fired parts using cutting, turning, and milling operations (green state), or on fired parts by grinding (dense state). Over the past 30 years, subtractive methods have benefited from innovations in computer-aided design and manufacturing (CAD/CAM) integrated with sophisticated machine tools. Components previously only produced by journeymen machinists and craftsmen are now routinely manufactured by CAM programmers and machine operators. With computer-based technology, intricately shaped parts are now available that previously were considered improbable. Subtractive manufacturing is remarkably flexible and ideal for short production runs and rapid-prototyping [631]. Nowadays, almost all ceramic orthopaedic implants are produced using this method, and it is also extensively used to produce custom dental implants [632]. However, there are drawbacks and limitations. By its very nature, subtractive methods involve waste of raw materials, tooling, energy, labor, and capital equipment. Furthermore, there are constraints to the complexity and feature sizes that can be addressed with this process. Intricate 3-dimensional scaffolds, with sub-millimeter and hidden internal features are nearly impossible, although creative processing, machining, and bonding techniques have been employed to produce prototypes [633,634].

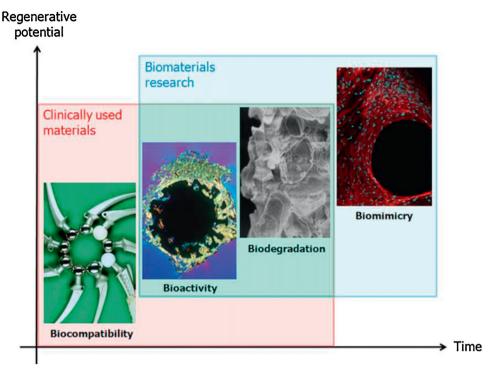


Fig. 18. Evolution of biomaterials. Today's clinical materials will progress from being bioinert and biocompatible to bioactive and biodegradable, to ultimately mimicking or stimulating cellular function [629]. Reprinted with permission.

Near-net-shape fabrication was the next logical progression in ceramic manufacturing. While, it too, involves forming of components by the aforementioned methods, in this case, parts are produced to achieve essentially their final size subsequent to firing. Very little intermediate green machining or final grinding are required, and usually only to remove mold flashings. However, achieving net shape dimensions necessitates tighter process controls and more intensively designed and costly forming tools. This method obviously reduces waste and concomitant costs. Intricately shaped components (particularly produced using injection molding or gel casting) are more common; but they still lack extremely fine or hidden features and close tolerances.

Additive manufacturing is the latest evolutionary development in ceramic production. Also known as solid freeform fabrication (SFF), this method has been exclusively used for prototypes and models until recently. It involves producing solid, porous, or functional gradient components from computer models without the use of special fixtures or tooling. Parts are manufactured by the successive layer-by-layer build-up of ceramic powders. Components can be highly complex, yet still possess fine features. As reviewed by Travitzsky et al., multiple SFF processes have been developed. While a number of SFF methods have been examined for bioceramics, those with particular relevance include 3D printing (3DP), selective laser sintering (SLS), and stereolithography (SLA) [635]. Given in the following paragraphs are brief discussions on each technique along with selected examples.

Three dimensional printing—3DP is perhaps the most widely known method for SFF, and is based on ink-jet printing developed at the Massachusetts Institute of Technology (MIT) in the 1990s [636]. Two deposition methods are employed—direct and indirect. Using the direct method, ceramic powder is dispersed within a liquid binder or wax, and then each layer is successively thermally printed using an injection nozzle. Components having intricate cavities and blind features with excellent surface finish are easily and rapidly formed. However, achieving high-solids content and controlling the rheology of the powder-ink mixture are key challenges.

Effective demonstration of this process was recently performed for a Y- TZP ceramic dental crown where the occlusal surfaces were precisely replicated from the computer model [637]. Using the indirect method, components are built by distributing a liquid binder onto successive powder layers. There is no need for admixing powder and ink in this method. Conversely, fine powders are difficult to spread evenly across the work surface, resulting in poor feature definition and lower density. Coarser powders are preferred. Both methods require post-forming sintering to achieve final density. 3DP has been effectively used for construction of hydroxyapatite bone scaffolds [638].

Selective laser sintering-SLS is a process in which particles within a powder bed are fused by the application of a precision laser. Here too, there are two methods-direct and indirect. The direct method relies on localized melting between ceramic particles. A low temperature liquid phase is a prerequisite, which limits the number of ceramic materials that can be employed. High thermal gradients can also result in component distortion and cracking. A polymeric binder is employed in the indirect method. Fusion occurs via melting of the binder between adjacent particles followed by coalescence. Subsequent sintering is required to fully densify the as-formed parts. In either method, the density of the components is directly related to the density of the powder bed. Higher powder bed densities result in higher density parts. Using the direct SLS method, Feng et al. added 45S5 glass into β-tricalcium phosphate (which provided the necessary lowtemperature liquid) to form complex bone scaffolds [639]. These were subsequently conventionally sintered to near full-density, yielding excellent mechanical properties and favorable biocompat-

Stereolithography—in the SLA method, ceramic powders are dispersed into a photo-sensitive monomer resin. This liquid mixture is then selectively solidified layer by layer upon exposure to an ultraviolet laser. Afterwards, formed components are subjected to general irradiation or a thermal treatment to cure the remaining monomers. Following curing, the monomers are removed in a burn-

out cycle and components are densified by conventional sintering. Similar to the other SFF methods, SLA is capable of forming intricate shapes with hidden features. However, it is not as capable of reproducing computer models as 3DP due to wider photo line widths and flow of uncured resin during part formation. These limitations result in rougher surface finishes with stepped features. Nevertheless, custom-made anatomical ceramic implants, such as hydroxyapatite scaffolds, have been produced using SLA techniques [640], and this method remains broadly applicable for a number of non-ceramic biomedical devices [641].

The one ceramic implant common to all three SFF methods is an orthophosphate bone scaffold. Indeed, SFF technologies offer significant benefits over conventional manufacturing for these devices, and are ideally suited for building complex, patient-customized prostheses [642]. Present-day SFF equipment can readily create sub-millimeter features and internal pore structures which effectively support neovascularization and osteogenesis. However, with the advent of higher precision equipment, building of micron and submicron features is on the horizon [643]. Continued development of these technologies will pave the way for their use in many reconstructive or arthroplastic procedures, including trauma [644], oral [645], cranio-maxillofacial [646], and spinal [647] surgeries, and ultimately total joint replacement [648].

4.3. Testing methods

Currently, validation of new ceramic orthopaedic prostheses requires conducting a series of independent tests on both samples and components. For example, consider the regulatory requirements for a new ceramic-on-polyethylene total hip arthroplasty system using a titanium hip stem [649]. The ceramic's basic strength is determined by flexural testing of small parallelepiped specimens. Compression burst testing is conducted on actual ceramic femoral heads against mock titanium stems. Separate fatigue testing of the femoral heads is also performed. Corrosion between the femoral heads and hip stems is determined using another test. Additional testing is conducted to assess wear in a hip simulator. While each test is performed in accordance with international standards, the results are independently reported and only weakly correlated. Provided each test meets its successful criteria, the total hip system is considered acceptable for market clearance. Little consideration is given to interactive effects. However, the hip system's real world behavior may differ considerably from these independent evaluations. Multiple physical effects interacting simultaneously might result in unpredicted early failures. As an example, wear, fatigue, and hydrothermal ageing are now known to be the main degradation mechanisms of ceramics used for joint prostheses. All three occur simultaneously in vivo, and coupling them in preclinical testing may correctly simulate the complex conditions encountered within the body. Yet, to date, only a few recent papers have highlighted multi-failure mode interactions [248,650]. Realization of this problem has led to the use of multiphysics simulation and experimental verification testing. Multiphysics simulation is a recent approach to computer modeling which simultaneously evaluates the effects of multiple environmental and operating parameters on product performance and reliability [651]. It has been adopted in the aerospace and automotive industries with considerable success [652,653], but is only now beginning to be used for orthopaedics [654]. Recent marked increases in computational power have enabled complex material and system simulations which heretofore have not been possible. Sophisticated multiphysics software is now included with most finite element modeling and CAD systems. These new tools provide biomedical engineers with the ability to in silco test and evaluate concepts and conditions simultaneously. It decreases design errors and costs, and minimizes reliance on multiple prototypes and test specimens. However, it does not obviate verification testing. Conversely, it is used to discover the physicochemical and mechanical tests that are most important or overlooked. As described earlier in this review, orthopaedics in general and joint arthroplasty in particular have had products successfully clear independent *in vitro* testing, only to later fail *in vivo*. Adoption and utilization of multiphysics simulation and experimental testing is an important advancement to eliminate the trial and error failures of the past.

5. Conclusions

In the present review, rather than discussing each ceramic material sequentially, a brief history of their evolution was provided, followed by detailed comparisons of respective properties and performance. This was done in order to highlight and contrast differences, advantages, and limitations of these bioceramics. Much of the early work in these materials was directed at total hip arthroplasty. Improvements in wear performance via introduction of Al₂O₃ and osseointegration of hip stems with the advent of HAp coatings are notable milestones. However, other wellintentioned improvements were conducted using trial and error methods, many times with insufficient laboratory and clinical data to support the safety and efficacy of new devices. Notable failures include femoral heads produced using DLC-coated titanium and Y-TZP. Both resulted in considerable patient harm prior to implant removals and the recall of these devices. With the possible exception of OxZr, other hard ceramic surface coatings appear to be headed towards a similar fate, at least from an engineering standpoint, principally due to deposition defects and an inability to withstand third-body wear. Despite improvements in the mechanical properties of ceramics, particularly with the introduction of ZTA, AMC, and Si₃N₄, an unwarranted concern over catastrophic fracture has been an impediment to their adoption by clinicians. Unfortunately, this resulted in further patient suffering associated with failures of metal-on-metal devices. Yet, with exceptions of Al₂O₃ and HAp, both of which have 30+ years of successful clinical history, concerns over newer bioceramics persist because their arthroplastic resumes are no greater than about 10 years. However, unlike prior generations of ceramics, currently available materials have been extensively vetted in vitro and in vivo, using both standardized and exploratory test methods that are more clinically relevant today than ever before. Regulatory agencies have also commensurately increased their acceptance criteria to ensure the safety and efficacy of devices prior to their release. In turn, the requirements imposed on device manufacturers continue to rise as well. In a recent instructional review by Rajpura et al. [655], the perfect total hip arthroplasty prosthesis needs to be constructed of articulation materials that are reasonably hard, scratch resistant, mechanically tough and strong with excellent in vivo biocompatibility and phase stability. It must possess a low friction interface that produces no noise and essentially no wear, with no immunological response to wear debris. In addition, it must have a large femoral head to prevent dislocation, and a head-neck interface that is not susceptible to fretting or electrochemical corrosion. However, while these are demanding requirements, some of which are still unmet, they are likely insufficient for the next generation of bioceramics. No longer will it simply be acceptable to focus on producing biocompatible or bioinert components. In fact, as discussed within this review, no materials are perfectly bioinert. Even the most established bioceramics are subject to chemical or phase instabilities-some of which are detrimental, while others may be beneficial. Indeed, instead of producing bioceramics that are wholly resistant to change, the next generation of ceramics must interact with human tissues and cells. Certainly reducing friction, suppressing noise, and eliminating wear debris are important objectives, but of equivalent importance is eradication of peri-operative and latent infections. While continued improvements in the strength and toughness of bioceramics are desirable, designing them as composite structures to interdigitate with adjacent bone is also needed. The next generation of "smart bioceramics" needs to be customized for specific arthroplasty applications. They should be bioactive, possessing engineered topography and functionalized surface chemistry to selectively and positively interact with adjacent tissue or bone. Only by doing so, will the dream of lifetime implants be realized. Achieving this level of performance for a new generation of ceramics and ceramic coatings will demand the best cooperative efforts from clinicians, manufacturers, and regulatory agencies.

Disclosure

Bryan J. McEntire and B. Sonny Bal are principals of Amedica Corporation, a silicon nitride orthopaedic device manufacturer. None of the other authors have a financial or proprietary interest in the subject matter or materials discussed.

Acknowledgements

Appreciation is expressed to Alan Lakshminarayanan, Ryan Bock, and Erin Jones for their support in literature searches, manuscript organization, and helpful discussions.

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